

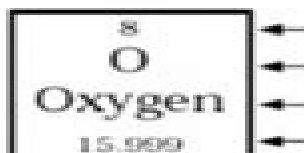
Protons, Neutrons, and Electrons Practice Worksheet

Calculating the number of each particle in an atom:

Protons = Atomic Number

Electrons = Protons

Neutrons = Atomic Mass – Atomic Number OR Big # - Small



Use the periodic table to find the numbers of protons, neutrons, and electrons for atoms of the following elements.

Name of Element	Element Symbol	Mass Number	Atomic Number	Protons	Neutrons	Electrons
Boron	B	11	5	5	6	5
Sodium	Na	24	11	11	13	11
Yttrium	Y	89	39	39	50	39
Copper	Cu	64	29	29	35	29
Technetium	Tc	98	43	43	55	43
Lead	Pb	207	82	82	125	82
Thallium	Tl	204	81	81	123	81
Hydrogen	H	1	1	1	0	1
Carbon	C	12	6	6	6	6
Nitrogen	N	14	7	7	7	7
Barium	Ba	137	56	56	81	56
Calcium	Ca	40	20	20	20	20
Silicon	Si	28	14	14	14	14
Argon	Ar	40	18	18	32	18
Magnesium	Mg	24	12	12	12	12

Chemfiesta Answers Protons Neutrons

WJ Hussar



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