

# LIMITING REACTANT STOICHIOMETRY WORKSHEET

Name _____	Section _____	Date _____
<b>Limiting Reactants</b>		
<b>Directions:</b> Determine the limiting reactant in each problem.		
1) In $2\text{NaOH} + \text{MgCl}_2 \rightarrow \text{Mg(OH)}_2 + 2\text{NaCl}$ , 5.818 grams of NaOH react with 2.712 grams of $\text{MgCl}_2$ .		
2) 0.0718 grams of HCl react with 0.0002 grams of $\text{Ca(OH)}_2$ in the reaction $2\text{HCl} + \text{Ca(OH)}_2 \rightarrow 2\text{H}_2\text{O} + \text{CaCl}_2$ .		
3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2008 grams of Hydrogen.		

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1) In $2\text{NaOH} + \text{MgCl}_2 \rightarrow \text{Mg(OH)}_2 + 2\text{NaCl}$ , 5.818 grams of NaOH react with 2.712 grams of $\text{MgCl}_2$ .		
2) 0.0718 grams of HCl react with 0.0002 grams of $\text{Ca(OH)}_2$ in the reaction $2\text{HCl} + \text{Ca(OH)}_2 \rightarrow 2\text{H}_2\text{O} + \text{CaCl}_2$ .		
3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2008 grams of Hydrogen.		
4) In the reaction $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$ , 10.000 grams of $\text{O}_2$ react with 0.908 grams of $\text{C}_2\text{H}_2$ .		
5) In the reaction $2\text{Al(OH)}_3 + 3\text{H}_2\text{CO}_3 \rightarrow \text{Al}_2(\text{CO}_3)_3 + 6\text{H}_2\text{O}$ , 1.18 grams of $\text{Al(OH)}_3$ and 1.012 grams of $\text{H}_2\text{CO}_3$ combine.		
6) 0.838 grams of sodium and 0.7021 grams of sodium nitrate react in the following process: $10\text{Na} + 2\text{NaNO}_3 \rightarrow 6\text{Na}_2\text{O} + \text{N}_2$ .		
7) 0.0700 grams of $\text{AlBr}_3$ react with 1.201 grams of $\text{H}_2\text{SO}_4$ to produce products in the following reaction: $2\text{AlBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 6\text{HBr}$ .		
8) When $\text{K}_3\text{PO}_4$ and NaOH react in $\text{K}_3\text{PO}_4 + 3\text{NaOH} \rightarrow 3\text{KOH} + \text{Na}_3\text{PO}_4$ , 5.41 grams of NaOH react with 3.88 grams $\text{K}_3\text{PO}_4$ .		
9) In $\text{Pt}_2(\text{NO})_2 + \text{H}_2\text{S} \rightarrow \text{PtS} + \text{HNO}_3$ , 10.02 grams of $\text{Pt}_2(\text{NO})_2$ react with 7.98 grams of $\text{H}_2\text{S}$ .		
10) 0.4000 grams of $\text{H}_2\text{SO}_4$ combine with 0.2480 grams of $\text{Ba(OH)}_2$ in the reaction $2\text{H}_2\text{SO}_4 + \text{Ba(OH)}_2 \rightarrow \text{Ba(SO}_4)_2 + 4\text{H}_2\text{O}$ .		

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A CHEMISTRY WORKSHEET

10 PROBLEMS

WORKED ANSWER KEY

# Chemfiesta Stoichiometry Limiting Reagents Practice Answers

**J Dewey**



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