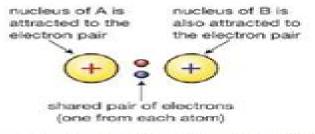
CHAPTER 8: COVALENT BONDING

1.44 Know that a covalent bond is formed between atoms by the sharing of a pair of electrons

 Covalent Bond: bond between non-metal and non-metal formed by a shared pair of electrons between the two atoms. It occurs between atoms of non-metal elements.

1.45 Understand covalent bonds in terms of electrostatic attractions

In a covalent bond, a pair of electrons is shared between two atoms. What holds the
atoms together is the strong electrostatic attraction between the nuclei (positively
charged) of the atoms that make up the bond, and the shared pair of electrons (negatively
charged).



1.46 Understand how to use dot-and-cross diagrams to represent covalent bonds in:

 Diatomic molecules, including hydrogen, oxygen, nitrogen, halogens and hydrogen halides



Chemhnhs Chapter 8 Covalent Bonding

SJ Ball

Chemhnhs Chapter 8 Covalent Bonding:

The book delves into Chemhnhs Chapter 8 Covalent Bonding. Chemhnhs Chapter 8 Covalent Bonding is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemhnhs Chapter 8 Covalent Bonding, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 1: Introduction to Chemhnhs Chapter 8 Covalent Bonding
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 - Chapter 4: Chemhnhs Chapter 8 Covalent Bonding in Specific Contexts
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