



Chapter 3: Atoms And Molecules

Science Worksheet

Standard: 9th

Subject: Science

Student Name: _____

Q1. In a chemical reaction 10.6g of sodium carbonate reacted with 12g of ethanoic acid. The products were 4.4g carbon dioxide, 1.8g of water and sodium ethanoate. The mass of sodium ethanoate formed is

1. 16.4g
2. 0.16g
3. 24g
4. 8.2g

Q2. When two molecules of hydrogen react with one molecule of oxygen, the mass of reactants is 36, what is the mass of products:

1. 18
2. 36
3. 9
4. 32

Q3. Barium chloride reacts with sodium sulphate to form Barium sulphate and sodium chloride. Then, according to the law of conservation of mass:

1. The total mass of reactants, barium chloride, and sodium sulphate taken is greater than the total mass of products, barium sulphate, sodium chloride formed
2. The total mass of reactants, barium chloride, and sodium sulphate is less than the total mass of products, barium sulphate, and sodium chloride
3. The total mass of the reactants is equal to the total mass of products
4. None of the above

Q4. 'All matter is composed of very small particles called Parmanu' was first suggested by:

1. John Dalton
2. J.J. Thomson
3. Kanad
4. William J. Crooke

Q5. A particle X has 17 protons, 18 neutrons and 18 electrons. This particle is most likely to be:

1. A cation.



Chemical Arithmetic Atoms And Molecules Answers

Oscar Oldberg



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Course in pharmaceutical and chemical arithmetic Julius William Sturmer,1917 **Chemical Arithmetic and Calculation of Furnace Charges** Regis Chauvenet,1912 **Problems in Chemical Arithmetic** Ebenezer John Cox,1891

Chemistry For Aieee , *Study Guide and Solutions Manual to accompany Basic Concepts of Chemistry, 9th Edition* Leo J. Malone,2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone s Basic Concepts of Chemistry

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emphasize or omit certain clearly delineated sections The mission of the text has evolved in response to the increased diversity of students and the emphasis on outcomes assessment *General Chemistry* James E. Brady, Gerard E. Humiston, 1980 **Chemistry** Frank Brescia, 1978 **American Druggists' Circular and Chemical Gazette** , 1921 A
Cyclopedia of Education Paul Monroe, 1911 **Science for the School and Family: Chemistry** Worthington Hooker, 1876
Student's Guide to Chemistry; a Modern Introduction David Brooks, 1974

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Table of Contents Chemical Arithmetic Atoms And Molecules Answers

1. Understanding the eBook Chemical Arithmetic Atoms And Molecules Answers
 - The Rise of Digital Reading Chemical Arithmetic Atoms And Molecules Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Arithmetic Atoms And Molecules Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Arithmetic Atoms And Molecules Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Arithmetic Atoms And Molecules Answers
 - Personalized Recommendations
 - Chemical Arithmetic Atoms And Molecules Answers User Reviews and Ratings
 - Chemical Arithmetic Atoms And Molecules Answers and Bestseller Lists

5. Accessing Chemical Arithmetic Atoms And Molecules Answers Free and Paid eBooks
 - Chemical Arithmetic Atoms And Molecules Answers Public Domain eBooks
 - Chemical Arithmetic Atoms And Molecules Answers eBook Subscription Services
 - Chemical Arithmetic Atoms And Molecules Answers Budget-Friendly Options
6. Navigating Chemical Arithmetic Atoms And Molecules Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Arithmetic Atoms And Molecules Answers Compatibility with Devices
 - Chemical Arithmetic Atoms And Molecules Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Arithmetic Atoms And Molecules Answers
 - Highlighting and Note-Taking Chemical Arithmetic Atoms And Molecules Answers
 - Interactive Elements Chemical Arithmetic Atoms And Molecules Answers
8. Staying Engaged with Chemical Arithmetic Atoms And Molecules Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Arithmetic Atoms And Molecules Answers
9. Balancing eBooks and Physical Books Chemical Arithmetic Atoms And Molecules Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Arithmetic Atoms And Molecules Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Arithmetic Atoms And Molecules Answers
 - Setting Reading Goals Chemical Arithmetic Atoms And Molecules Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Arithmetic Atoms And Molecules Answers
 - Fact-Checking eBook Content of Chemical Arithmetic Atoms And Molecules Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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