- Atoms form molecules by sharing electrons and those are known as covalently bonded molecules. As same, by donating electrons to non-metals from metals the sonic compounds are forms.
 - a) Group the following compounds ionic or covalent.

Carbon dioxide Magnesium Aluminium oxide water sulfur trioxide

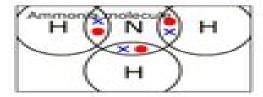
Potassium iodide methane ethanol oxygen ammonium chloride ammonia gas

Diamond copper nitrate silicon dioxide sodium chloride hydrogen chloride gas

Covalent compounds	tenic compounds
Carbon dicoide	Aluminium oxide
Viriabed	Potassium iodide
Sulfur transide	Copper nitrate
Mothania	Sodium chloride
Ethanol	
Owygen	
Ammonia gas	5
Silicon dioxide	
Hydrogen gas	

- b). Simple molecular compounds can be formed by sharing electrons only. Give 3 examples for simple molecular compounds
- c. Draw dot cross diagrams to show the bonding in: i) hydrogen chloride molecule ii) ammonia molecule





- d) Explain why the bond in a nitrogen molecule is called a triple bond.
 - Two nitrogen atoms share three electrons from each atom to form three covalent bonds.
- e). How many electrons are need to form a covalent bond?

Two electrons / a pair of electrons

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