Bonding Test

Do Not Write on This Test!!! You may use a periodic table.

Multiple Choice-Identify the choice that best completes the statement or answers the question.

- Electrons involved in bonding between atoms are
 - a. valence electrons b. inside the nucleus
 - c. closest to the nucleus
 d. positively charged
- Each family in the periodic table has its own characteristic properties based on the number of
 - a, neutrons b. valence electrons
 - d ions c. protons
- What is the areatest number of valence electrons an atom can have? With the exception of helium.

d. 12

- b. 3 c. 8 0.2
- 4) If atoms of a halogen nonmetal (Group 17) gains one electron, the atoms then have b. 7 valence electrons
 - a, no valence electrons
 - c. 8 valence electrons d. 17 valence electrons
- When an atom loses an electron, it becomes a
 - a, positive ion b. negative ion
 - c. neutral ion d. neutral atom
- An ionic bond is the attraction between
 - a, similarly charged ions b. oppositely charged ions
 - d. neutral atoms c. neutral ions
- The element boron is directly above aluminum on the periodic table. Which statement about boron is true?
 - a. Boron is in the same period as aluminum
 - b. Boron is in the same group as aluminum and has 5 valence electrons
 - c. Boron has 5 valence electrons and is in the same period as aluminum
 - d. Boron is in the same group as aluminum and has 3 valence electrons
- Magnesium bromide is an ionic compound with the chemical formula MgBr2. What does the "2" fell you?
 - a. Bromide has a 2⁻ charae.
 - b. There are two magnesium ions to every bromide ion.
 - There are two bromide ions for every magnesium ion.
 - d. Bromide has a 2+ charge.
- 9) What is the chemical name for the compound with the formula Na:5?
 - a. sodium fluoride b. magnesium sulfide
 d. sodium sulfide
 - c. lithium oxide
- 10) In the chemical formula for an ionic compound, which item is written first?
 - a, positive ion b, negative ion
 - c. subscript d, charae

Chemical Bonding Multiple Choice Test

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