

Chemical Bonding Worksheet

Ionic Bond
 Covalent Bond
 Metallic Bond

between a Metal and Non-Metal
 between a Non-Metal and Non-Metal
 between a Metal and Metal

Compound	Name	Oxidation states each atom	of	Bond Type
NO ₂	nitrogen dioxide	N = +4 O = -2		covalent
NaCl	sodium chloride	Na = +1 Cl = -1		ionic
SO ₂	sulfur dioxide	S = +4 O = -2		covalent
PO ₄ ³⁻	phosphate anion	P = +5 O = -2		covalent
MgBr ₂	magnesium bromide	Mg = +2 Br = -1		ionic
CaO	calcium oxide	Ca = +2 O = -2		ionic
H ₂ O	water	H = +1 O = -2		covalent
K ₂ O	potassium oxide	K = +1 O = -2		ionic
Cu-Zn alloy	brass	Cu = 0 Zn = 0		metallic
O ₂	oxygen gas	O = 0		covalent
CuCl ₂	copper(II) chloride	Cu = +2 Cl = -1		ionic
NO ₂ ⁻	nitrite anion	N = +3 O = -2		covalent
TiO ₂	titanium (IV) oxide	Ti = +4 O = -2		ionic
HF	hydrofluoric acid	H = +1 F = -1		covalent
Rb ₂ S	rubidium sulfide	Rb = +1 S = -2		ionic
Au-Pt mixture	white gold	Au = 0 Pt = 0		metallic
Fe ₂ O ₃	iron (III) oxide	Fe = +3 O = -2		ionic
C ₆ H ₁₂ O ₆	glucose	C = +4 H = -1 O = -2		covalent

Chemical Bonding Worksheet 13 Answer Key

**Samuel N. Namowitz, Nancy E.
Spaulding**



Chemical Bonding Worksheet 13 Answer Key:

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