Bleicky

Lab Report

Chemical Reactions

It's time to complete your Lab Report. Save the lab to your computer with the correct unit number, lab name, and your name at the end of the file name (e.g., U3_Lab_ChemicalReactions_Alice_Jones.doc).

Introduction

1. What was the purpose of the experiment?

Type your answer here: the purpose of this experiment is to

How can you identify the gas produced in a chemical reaction without observing the atoms and molecules directly? Explain.

Type your answer here

3. Write a hypothesis based on observations and scientific principles.

Type your answer here.

Experimental Methods

1. What tools or materials did you use to collect your data?

Type your answer here

2. Describe the procedure that you followed to conduct your experiment.

Type your answer here:

Data and Observations

Record your observations.

Type your answer here

Conclusions

What conclusions can you draw about the identities of the gases produced in each chemical reaction? Write an evidence-based claim.

Type your answer here:

- Based on your observations, which equation best models the reaction of vinegar (C₂H_{*}O₂) and baking soda (NaHCO₃)? Explain.
 - (a) $C_2H_4O_2 \cdot NaHCO_3 \rightarrow NaC_2H_3O_2 \cdot H_2O \cdot CO_2$
 - (b) $C_2H_4O_2 * NaHCO_3 = NaC_2H_3O_2 * H_2O * O_2$

Chemical Lab Report Chemical Reactions And Equations

RC Schank

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School Science and Mathematics ,1909

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A Laboratory Book of Elementary Organic Chemistry Alexander Lowy, Wilmer E. Baldwin, 1926 Merck's Report
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