#### Six Types of Chemical Reaction Worksheet Answers

Balance the following reactions and indicate which of the six types of chemical reaction are being represented:

a) 2 NaBr + 1 Ca(OH)₂ → 1 CaBr₂ + 2 NaOH

Type of reaction: double displacement

b) 2 NH<sub>3</sub>+ 1 H<sub>2</sub>SO<sub>4</sub> → 1 (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>

Type of reaction: synthesis

c) 4 C<sub>8</sub>H<sub>9</sub>O + 27 O<sub>2</sub> → 20 CO<sub>2</sub> + 18 H<sub>2</sub>O

Type of reaction: combustion

d) 3 Pb + 2 H<sub>3</sub>PO<sub>4</sub> → 3 H<sub>2</sub> + 1 Pb<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>

Type of reaction: single displacement

e) 1 Li<sub>2</sub>N + 3 NH<sub>4</sub>NO<sub>3</sub> → 3 LiNO<sub>3</sub> + 1 (NH<sub>4</sub>)<sub>3</sub>N

Type of reaction: double displacement

f) 3 HBr + 1 Al(OH)<sub>3</sub> → 3 H<sub>2</sub>O + 1 AlBr<sub>3</sub>

Type of reaction: acid-base (double displacement)

g) Na<sub>3</sub>PO<sub>4</sub> + 3 KOH → 3 NaOH + K<sub>3</sub>PO<sub>4</sub>

g) Najrou + 3 Non -> 3 Naon + Njrou

h) MgCl₂ + Li₂CO₃ → MgCO₃ + 2 LiCl

i) C<sub>8</sub>H<sub>12</sub> + 9 O<sub>2</sub> → 6 CO<sub>2</sub> + 6 H<sub>2</sub>O

i) Pb + FeSO<sub>4</sub> → PbSO<sub>4</sub> + Fe

k) CaCO<sub>3</sub> → CaO + CO<sub>2</sub>

P<sub>4</sub> + 3 O<sub>2</sub> → 2 P<sub>2</sub>O<sub>3</sub>

m) 2 RbNO<sub>3</sub> + BeF<sub>2</sub> → Be(NO<sub>3</sub>)<sub>2</sub> + 2 RbF

n) 2 AgNO<sub>3</sub> + Cu → Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag

O) C<sub>3</sub>H<sub>8</sub>O + 4 O<sub>2</sub> → 3 CO<sub>2</sub> + 3 H<sub>2</sub>O

p) 2 CsHs + Fe → Fe(CsHs)<sub>2</sub>

q) SeCl<sub>0</sub> + O<sub>2</sub> → SeO<sub>2</sub> + 3Cl<sub>2</sub>

r) 2 Mgl<sub>2</sub> + Mn(SO<sub>3</sub>)<sub>2</sub> → 2 MgSO<sub>3</sub> + Mnl<sub>4</sub>

s) O<sub>2</sub> → O + O<sub>2</sub>

t) 2 NO<sub>2</sub> → 2 O<sub>2</sub> + N<sub>2</sub>

Reaction Type \_\_double displacement\_\_\_\_

Reaction Type double displacement

Reaction Type \_\_\_\_complete combustion\_\_\_\_

Reaction Type \_\_\_\_single displacement\_\_\_\_

Reaction Type \_\_\_\_decomposition\_\_\_\_\_

Reaction Type \_\_\_synthesis\_\_\_\_\_

Reaction Type \_\_\_\_double displacement\_\_\_\_

Reaction Type \_\_\_single displacement\_\_\_\_

Reaction Type \_\_\_\_complete combustion\_\_\_\_

Reaction Type \_\_\_synthesis\_\_\_\_\_

Reaction Type \_\_\_\_single displacement\_\_\_\_\_

Reaction Type \_\_\_\_double displacement\_\_\_\_

Reaction Type \_\_\_\_decomposition\_\_\_\_\_

Reaction Type \_\_\_\_\_decomposition\_\_\_\_\_

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**Arshad Iqbal** 

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and a commitment to fostering a positive classroom environment I am also dedicated to ongoing professional growth and development to enhance my teaching practices Why do you think you would be a good fit for our school district Answer I am drawn to your school district s commitment to mention specific values or initiatives such as inclusivity personalized learning or community involvement My teaching philosophy aligns with these values and I am eager to contribute my skills passion for education and dedication to student success within your learning community These answers can be customized based on your teaching experiences educational philosophy and the specific requirements of the teaching position you are applying for

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