

11. What are the correct coefficients when this chemical equation is balanced?



- a. 4, 2, 7
- b. 1, 1, 1
- c. 2, 3, 4
- d. 1, 3, 2

12. Identify the type of reaction shown by this chemical equation.



- a. single replacement
- b. synthesis
- c. double replacement
- d. decomposition

13. Identify the type of chemical reaction demonstrated by this equation.



- a. decomposition
- b. synthesis
- c. combustion
- d. single replacement

14. What is the probable product of a double-replacement reaction?

- a. A new compound and the replaced metal
- b. A new compound and the replaced nonmetal
- c. Two different compounds
- d. A single compound

15. Which of the following statements about a catalyst is true?

- a. A catalyst can initiate a reaction.
- b. A catalyst can accelerate a reaction.
- c. A catalyst can be consumed during a reaction.
- d. A catalyst can be changed during a reaction.

16. Which of the following factors does NOT affect the rate of the reaction?

- a. the amount of the reactants
- b. the physical state of the reactants
- c. the size of the container used
- d. temperature

17. A(n) _____ is a substance that slows down the rate of a reaction.

- a. catalyst
- b. inhibitor
- c. reactant
- d. product

18. A(n) _____ lowers the activation energy required for a reaction to take place, thus increasing its rate.

- a. catalyst
- b. inhibitor
- c. reactant
- d. product

19. A catalyst requires _____ to be effective.

- a. only enough energy
- b. favorable orientation
- c. enough energy and favorable orientation
- d. a reaction mechanism

Chemical Reaction Test Answers

Ensheng Dong



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