

## 11

## CHEMICAL REACTIONS

**SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)**

*This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.*

**► Writing Chemical Equations (pages 321–323)**

1. A chemical reaction occurs when one or more \_\_\_\_\_ change into one or more new substances called \_\_\_\_\_.
2. The arrow in a reaction means \_\_\_\_\_.
3. Is the following sentence true or false? When there are two or more reactants or products, they are separated by an arrow. \_\_\_\_\_
4. Write a word equation that describes the following reactions.
  - a. Acetylene reacts with oxygen to produce carbon dioxide and water.  
\_\_\_\_\_
  - b. When heated, mercury(II) oxide reacts to form mercury and oxygen.  
\_\_\_\_\_
5. What is a chemical equation?  
\_\_\_\_\_  
\_\_\_\_\_
6. A chemical reaction that shows only the formulas, but not the relative amounts of the reactants and products is a(n) \_\_\_\_\_.
7. Identify the reactant(s) and product(s) in the chemical equation  $\text{Li} + \text{Br}_2 \longrightarrow \text{LiBr}$ .
  - a. reactant(s) \_\_\_\_\_
  - b. product(s) \_\_\_\_\_
8. Circle the letter of each statement that is true about a catalyst.
  - a. A catalyst is the new material produced as a result of a chemical reaction.
  - b. A catalyst is not used up in a chemical reaction.
  - c. A catalyst adds heat to a chemical reaction.
  - d. A catalyst speeds up a chemical reaction.

# Chemical Reactions Ch 11 1 Describing Chemical Reactions Answers

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*Chaos Theory* Kais A. M. Al Naimee, 2018-03-28 With a good background in nonlinear dynamics chaos theory and applications the author of this leading book gives a systematic treatment of the basic principle of nonlinear dynamics in different fields The contributions from leading international scientists active in the field provide a comprehensive overview of our current level of background on chaos theory and applications in different sciences In addition they show overlap with the traditional field of control theory in scientific community **Prentice Hall**

**Chemistry**, 2000 *Making Ammonia* Benjamin Johnson, 2022-05-10 This Open Access book discusses the progress of science and the transfer of scientific knowledge to technological application It also identifies the factors necessary to achieve this progress Based on a case study of the physical chemist Fritz Haber s discovery of ammonia synthesis between 1903 and

1909 the book places Haber's work in historical and scientific physicochemical context. The scientific developments of the preceding century are framed in a way that emphasizes the confluence of knowledge needed for Haber's success. Against this background, Haber's work is presented in detail along with the indispensable contributions of his colleague the physical chemist Walter Nernst and their assistants. The detailed accounts of scientific advancement remind us of the physical basis on which our scientific theories and ideas are built. Without this reminder we often forget how complex and how beautiful achievements in science can be.

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