Chemical Reactions

Signs of Chemical Reactions

What is a chemical reaction?

Chemical Formulas

A chamical formula uses _ too lidentify the in a compound and the

H₂O water A worther excellentable is enable: up-of _____ atoms of

bonded to _____ atom of



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ġέ					atoms of	
			.and.		_atoms of	

Chemical Equations

A chemical equation uses to describe a

chemical reaction. The starting materials in a chemical reaction are . The resulting substances from a reaction

placed on the left of the amow, and the the right of the arrow.

C + O. - CO.

Key Vocabulary Terms

- double replacement reaction.
- chemical equation. chemical formula
- chemical reaction.
 - * PRINCESOR
- * compound decomposition reaction - synthesis reaction.

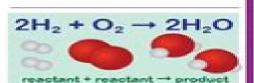
* autopern

- * effectivent * percellant
- single replacement reaction

Types of Chemical Reactions

Synthesis Reaction

Describe a synthesis reaction:





contribution top Foreign in mere adopted.

Decomposition Reaction

Describe a decomposition reaction:

2H	2O -	+ 2H ₂	+ 02
4	0	0	1
		00	
Fee	ectant	product +	product



Replacement Reaction

Single Replacement

Describe a single replacement reaction:

Zn + 2HCl → ZnCl₂+ H₂ reactant + reactant -+ product + product



Double Replacement

Describe a double replacement reaction:

NaCl + AgF → NaF + AgCl reactant + reactant -+ product + product



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