

Unit 11: Chapter 7 - Chemical Reactions Review Answer Key

Define:

1. **Reactant:** *A substance that undergoes change in a chemical reaction.*
2. **Product:** *New substances formed as a result of a chemical reaction.*
3. **Catalyst:** *A substance that affects the rate of a chemical reaction without being used up in the reaction.*
4. **Coefficient:** *Numbers that appear before a formula in a chemical equation to show the relative proportions of each reactant and product.*
5. **Subscript:** *A number written lower in a chemical formula indicating the amount of atoms present of an element (this number is not to be changed once written in chemical formula).*
6. **Law of Conservation of Mass:** *Matter cannot be created nor destroyed; only changed (atoms can be rearranged).*
7. **Synthesis reaction (include an example):** *A chemical reaction in which two or more substances react to form a single substance.* $(A + B \rightarrow AB)$
8. **Decomposition reaction (include an example):** *A chemical reaction in which a compound breaks down into two or more simpler substances.* $(AB \rightarrow A + B)$
9. **Single replacement (include an example):** *A chemical reaction in which one element takes the place of another element in a compound.* $(AB + C \rightarrow AC + B)$
10. **Double replacement (include an example):** *A chemical reaction in which two compounds exchange positive ions and form two new compounds.* $(AB + CD \rightarrow AD + CB)$
11. **Combustion (include an example):** *A chemical reaction in which a substance reacts rapidly with oxygen, often producing heat and light.* $(C_2H_2 + O_2 \rightarrow CO_2 + H_2O)$
12. **Endothermic reaction:** *A chemical reaction that absorbs energy from its surroundings.*
13. **Exothermic reaction:** *A chemical reaction that releases energy to its surroundings.*
14. **Chemical equation (include an example):** *A representation of a chemical reaction in which the reactants and products are expressed as formulas.*
15. **A balanced equation** is an equation in which the number of atoms on the reactants' side equals the number of atoms on the products' side of the chemical equation.
16. **Equilibrium:** *A state in which the forward and reverse paths of a physical or chemical change take place at the same rate.*

Chemical Reactions Review Answer

Benjamin Lillard



Chemical Reactions Review Answer:

Fundamentals of Environmental Chemistry, Third Edition Stanley E. Manahan, 2011-03-05 Written by an expert using the same approach that made the previous two editions so successful Fundamentals of Environmental Chemistry Third Edition expands the scope of book to include the strongly emerging areas broadly described as sustainability science and technology including green chemistry and industrial ecology The new edition includes Increased emphasis on the applied aspects of environmental chemistry Hot topics such as global warming and biomass energy Integration of green chemistry and sustainability concepts throughout the text More and updated questions and answers including some that require Internet research Lecturers Pack on CD ROM with solutions manual PowerPoint presentations and chapter figures available upon qualifying course adoptions The book provides a basic course in chemical science including the fundamentals of organic chemistry and biochemistry The author uses real life examples from environmental chemistry green chemistry and related areas while maintaining brevity and simplicity in his explanation of concepts Building on this foundation the book covers environmental chemistry broadly defined to include sustainability aspects green chemistry industrial ecology and related areas These chapters are organized around the five environmental spheres the hydrosphere atmosphere geosphere biosphere and the anthrosphere The last two chapters discuss analytical chemistry and its relevance to environmental chemistry Manahan s clear concise and readable style makes the information accessible regardless of the readers level of chemistry knowledge He demystifies the material for those who need the basics of chemical science for their trade profession or study curriculum as well as for readers who want to have an understanding of the fundamentals of sustainable chemistry in its crucial role in maintaining a livable planet Mining and Scientific Press and Pacific Electrical Review ,1916 Practical Druggist and Pharmaceutical Review of Reviews Benjamin Lillard,1911 *Digest; Review of Reviews Incorporating Literary Digest* ,1920 *Mining Industry and Review* ,1900 **Paint, Oil and Drug Review** ,1890 Review of Bacteriology, Protozoology and General Parasitology ,1916 The Review of Bacteriology, Protozoology, & General Parasitology ,1916 **The Commonwealth Forestry Review** ,1928 *Physical Review* ,1911 Vols for 1903 include Proceedings of the American Physical Society Medical Review ,1912 **New York Review of the Telegraph and Telephone and Electrical Journal** ,1899 *The Philosophical Review* Jacob Gould Schurman,James Edwin Creighton, Frank Thilly, Gustavus Watts Cunningham,1904 An international journal of general philosophy **American Review of Tuberculosis and Pulmonary Diseases** ,1924 Includes Abstracts section previously issued separately **The Physical Review** ,1897 **General Electric Review** General Electric Company,1921 **Weekly Medical Review** ,1883 **Brewers' Journal and Hop and Malt Trades' Review** ,1904 Practical Druggist and Pharmaceutical Review of Reviews ,1915 **The St. Louis Medical Review** ,1906

The book delves into Chemical Reactions Review Answer. Chemical Reactions Review Answer is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemical Reactions Review Answer, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemical Reactions Review Answer
- Chapter 2: Essential Elements of Chemical Reactions Review Answer
- Chapter 3: Chemical Reactions Review Answer in Everyday Life
- Chapter 4: Chemical Reactions Review Answer in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chemical Reactions Review Answer. The first chapter will explore what Chemical Reactions Review Answer is, why Chemical Reactions Review Answer is vital, and how to effectively learn about Chemical Reactions Review Answer.

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6. In chapter 5, this book will draw a conclusion about Chemical Reactions Review Answer. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chemical Reactions Review Answer.

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