

# Chemical Reactions Assignment

Name: \_\_\_\_\_

Period: \_\_\_\_\_

	I can balance chemical reactions
	I can classify chemical reactions and predict the products formed in chemical reactions
	I can predict the solubility of compounds and write net ionic equations

All questions must be completed on separate paper with all work shown. Attach your scanned work to the assignment on Teams.

1. Balance the following chemical reactions:

- $\text{C}_2\text{H}_5\text{OH} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{Al} + \text{Br}_2 \rightarrow \text{AlBr}_3$
- $\text{Al} + \text{H}_2\text{SO}_4 \rightarrow \text{H}_2 + \text{Al}_2(\text{SO}_4)_3$
- $\text{LiOH} + \text{Zn}(\text{NO}_3)_2 \rightarrow \text{Zn}(\text{OH})_2 + \text{LiNO}_3$

2. For the following sentences, write a balanced chemical equation:

- in the industrial synthesis of ammonia, known as the Haber process, the reaction of nitrogen gas and hydrogen gas is catalyzed by aluminum oxide to yield ammonia gas ( $\text{NH}_3$ ).
- Solid potassium chlorate decomposes to form solid potassium chloride and oxygen gas.

3. Classify the following reactions as either synthesis, decomposition, single replacement, double replacement, complete combustion or incomplete combustion by writing the name of the reaction.

- $\text{CaF}_2 + \text{Na} \rightarrow \text{NaF} + \text{Ca}$  \_\_\_\_\_
- $\text{F}_2 + \text{P}_4 \rightarrow \text{PF}_3$  \_\_\_\_\_
- $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2 + \text{H}_2$  \_\_\_\_\_
- $\text{C}_6\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  \_\_\_\_\_
- $\text{NaHCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{CO}_2 + \text{H}_2\text{O}$  \_\_\_\_\_
- $\text{H}_2\text{SO}_4 + \text{NH}_4\text{OH} \rightarrow \text{H}_2\text{O} + (\text{NH}_4)_2\text{SO}_4$  \_\_\_\_\_
- $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO} + \text{H}_2\text{O} + \text{C}$  \_\_\_\_\_

# Chemical Reactions Take Home Test

**Society of Chemical Industry (Great Britain)**



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**Essentials of Chemical Reaction Engineering** H. Scott Fogler, 2011 Accompanying DVD ROM contains many realistic interactive simulations

**The Hidden Curriculum—Faculty-Made Tests in Science** Sheila Tobias, Jacqueline Raphael, 2013-06-29 This resource manual for college level science instructors reevaluates the role of testing in their curricula and describes innovative techniques pioneered by other teachers part I examines the effects of the following on lower division courses changes in exam content format and environment revisions in grading practices student response colleague reaction the sharing of new practices with other interested professionals and more The book includes a comprehensive introduction faculty composed narratives commentaries by well known science educators and a visual index to 100 more refined innovations

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New York Supreme Court, Appellate Division- First Department , *The Johns Hopkins White Papers, 2007* Ross C. Donehower, 2007

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**Chemistry, Student Study Guide** John A. Olmsted, Gregory M. Williams, 2002-01-04 This third edition continues to innovate by providing students with an integrated and modern approach to the subject The text emphasizes the modern tools of chemistry while incorporating historical evidence and its unique molecular quantitative emphasis is further reinforced by an integrated media package developed by the authors Also of benefit is the just in time presentation of key content only providing details once they are needed While key topics and analytical techniques have been updated there is now an additional third chapter on chemical equilibrium The authors have also developed an expanded and more integrated problem solving emphasis that now incorporates a 4 step strategy throughout complete with text icons The whole is backed by a range of supplements including a new illustration program a tutorial CD interactive learningware an extensive Web CT component an instructor s resource CD and a solution CD

Fundamentals of Chemistry, Study Guide James E. Brady, John R. Holum, 1988-04-20 This Third

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### **Table of Contents Chemical Reactions Take Home Test**

1. Understanding the eBook Chemical Reactions Take Home Test
  - The Rise of Digital Reading Chemical Reactions Take Home Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Reactions Take Home Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemical Reactions Take Home Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Reactions Take Home Test
  - Personalized Recommendations
  - Chemical Reactions Take Home Test User Reviews and Ratings
  - Chemical Reactions Take Home Test and Bestseller Lists

5. Accessing Chemical Reactions Take Home Test Free and Paid eBooks
  - Chemical Reactions Take Home Test Public Domain eBooks
  - Chemical Reactions Take Home Test eBook Subscription Services
  - Chemical Reactions Take Home Test Budget-Friendly Options
6. Navigating Chemical Reactions Take Home Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemical Reactions Take Home Test Compatibility with Devices
  - Chemical Reactions Take Home Test Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemical Reactions Take Home Test
  - Highlighting and Note-Taking Chemical Reactions Take Home Test
  - Interactive Elements Chemical Reactions Take Home Test
8. Staying Engaged with Chemical Reactions Take Home Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemical Reactions Take Home Test
9. Balancing eBooks and Physical Books Chemical Reactions Take Home Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemical Reactions Take Home Test
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemical Reactions Take Home Test
  - Setting Reading Goals Chemical Reactions Take Home Test
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Reactions Take Home Test
  - Fact-Checking eBook Content of Chemical Reactions Take Home Test
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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