

Name _____

Date ____/____/____

ATOMS AND MOLECULES

Circle the letter of the correct answer.

1. A substance made up entirely of a single type of atom is which of the following?
a. Compound b. Enzyme c. Mixture d. Element
2. The smallest unit of any element is which of the following?
a. Atom b. Compound c. Molecule d. Enzyme
3. Different elements are chemically bonded together in _____.
a. Compounds b. Elements c. Mixtures d. Molecules
4. Fluorine is represented by which of the following chemical symbols? _____.
a. F b. Fe c. Fr d. Fl
5. The compound: $C_6H_{12}O_6$ contains how many atoms of oxygen?
a. 4 b. 6 c. 12 d. 24
6. Which of the following is a compound?
a. Oxygen b. Gold c. Sodium Chloride d. Sodium
7. Two or more atoms are bonded together are referred to as a(n) _____.
a. Atom b. Element c. Mixture d. Molecule
8. When a chemical bond is broken, what is released?
a. Water b. Energy c. Salt d. Oxygen
9. How many compounds are shown in the chemical equation: $C + O_2 \longrightarrow CO_2$
a. 1 b. 2 c. 3 d. 4
10. Which of the following terms refers to the smallest part of a compound that has all the properties of that compound?
a. Atom b. Cell c. Element d. Molecule
11. An atom which has 9 protons, 8 neutrons, and 9 electrons will have an atomic mass of how many amu?
a. 6 b. 8 c. 9 d. 17
12. An atom with an atomic number of 7 and a mass number of 8 is which element?
a. H b. N c. O d. P
13. Usually, the atom described in Problem 12 will have how many electrons?
a. 4 b. 7 c. 8 d. 15
14. An atom of Magnesium has how many protons?
a. 12 b. 20 c. 24 d. 25
15. When an atom becomes a compound, a _____ forms.
a. Bond b. Electron c. Neutron d. Oxygen

Chemical Arithmetic Worksheet Answers Atoms And Molecules 2

Michael Robinson, Mike Taylor



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