

# CHEMICAL BONDS

Name: \_\_\_\_\_

## WHAT IS A CHEMICAL BOND?

A bond is a force that holds atoms together and makes them function together as "one unit".



## TYPES OF BONDS



## ELECTRON CONFIG

Before bond

After bond

Before bond

After bond

## WHAT IS BONDING?

- Bonding energy is the amount of measured to break a bond.
- When atoms bond, it is because they can achieve the lowest possible energy by bonding. A bond will form if the bonded atoms is lower energy of the two separated.

## COVALENT BOND

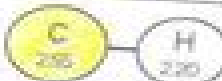
- A covalent bond is formed when each of the metal of two atoms.
- In general, nonmetal atoms react to form covalent bonds.

# ELECTRONEGATIVITY

Name: \_\_\_\_\_

Electronegativity is a measure of the tendency of an atom to attract a bonding pair of electrons. Scientists use the electronegativity difference to define the type of bond that is occurring between two atoms.

Electronegativity Difference = 0.35  
Nonpolar Covalent Bond



## NONPOLAR COVALENT BOND

- When both atoms are equally electronegative, the bond is called a nonpolar covalent bond. In a nonpolar covalent bond, the electronegativity difference is less than 0.5.

## POLAR COVALENT BOND

- When one atom in a covalent bond is more electronegative than the other, the atoms share the electrons.

## ELECTRONEGATIVITY TREND

ELECTRONEGATIVITY TREND																		Atomic Number		Group Number																									
1	2											3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																		
H 1.0	Li 0.9	Be 1.5											B 2.0	C 2.5	N 3.0	O 3.5	F 4.0	Ne 4.0	Na 0.9	Mg 1.2	Al 1.5	Si 1.8	P 2.1	S 2.5	Cl 3.0	Ar 3.0	K 0.8	Ca 1.0																	
Sc 1.3	Ti 1.3	V 1.6	Cr 1.6	Mn 1.5	Fe 1.8	Co 1.8	Ni 1.9	Cu 1.9	Zn 1.9	Ga 1.6	Ge 2.0	As 2.2	Se 2.4	Br 2.8	Kr 2.8	Rb 0.8	Sr 1.0	Y 1.2	Zr 1.4	Nb 1.6	Mo 2.2	Tc 2.1	Ru 2.2	Rh 2.2	Pd 2.2	Ag 1.9	Cd 1.7	In 1.8	Sn 1.8	Sb 2.0	Te 2.1	I 2.5	Xe 2.6												
La 1.0	Ce 1.1	Pr 1.1	Nd 1.1	Pm 1.1	Sm 1.2	Eu 1.2	Gd 1.2	Tb 1.2	Dy 1.2	Ho 1.2	Er 1.2	Tm 1.2	Yb 1.2	Lu 1.2	Hf 2.2	Ta 2.4	W 2.6	Re 2.5	Os 2.6	Ir 2.4	Pt 2.2	Au 2.4	Hg 2.0	Tl 1.8	Pb 2.0	Bi 2.0	Po 2.0	At 2.2	Fr 0.7	Ra 0.9	Ac 1.1	Th 1.3	Pa 1.3	U 1.3	Np 1.3	Pu 1.3	Am 1.3	Cm 1.3	Bk 1.3	Cf 1.3	Es 1.3	Fm 1.3	Md 1.3	No 1.3	Lr 1.3

# Chemical Bonding

# Doodle Diagrams



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**NARAYAN CHANGDER**



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web d there will not be any make up for lessons and activities missed during the student s absence 24 parent guardian s acknowledgement i have read and acknowledged the following a the school will give you a reply on this b the progress report of your child ward will reflect the total possible days of attendance and the number of days

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