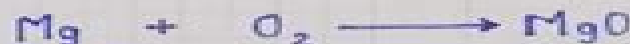


Chemical Reactions

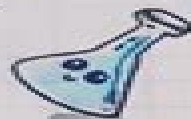
Chemical Equation

Reactants \longrightarrow Products
Left hand side Right hand side

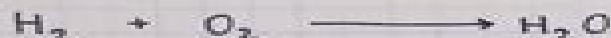
magnesium + Oxygen \longrightarrow magnesium oxide



BALANCING EQUATIONS



number of atoms of every element before and after reaction is same



Chemical equations are a way of showing chemical reactions.

LAW OF CONSERVATION OF MATTER
a chemical equation must be balanced.

Chemical reactions

COMBINATION

Two or more substance combine to form a single product.



DECOMPOSITION

Single reactant breaks up to form two or more products



DISPLACEMENT

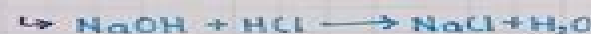
More reactive element removes the other element from the compound and itself forms the compound



- Chemical reactions can occur only:
 \hookrightarrow when the chemical change.
 Rusting of iron, Burning of coal.

DOUBLE DISPLACEMENT

Exchange of ions between the reactants take place.



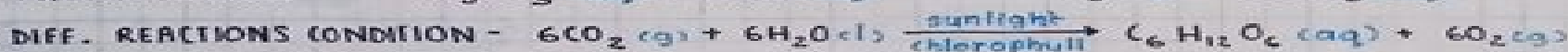
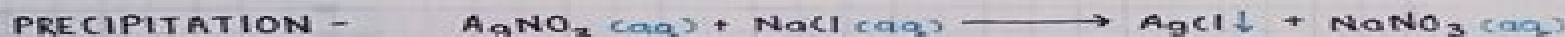
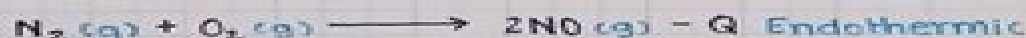
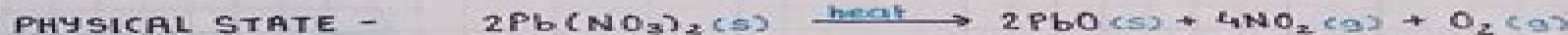
REDUCTION

gain of hydrogen or loss of oxygen.

redox

OXIDATION

gain of oxygen or loss of hydrogen.



Chemical Changes And Equations Teacher Notes

Joacim Rocklöv



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Catalogue of the Officers and Students of the College of New Jersey for College of New Jersey (Princeton, N.J.), Princeton University, 1902 **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents **Chemical Reactions, Materials and Particles** Gill Murphy, 2006-03 Specials are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age These books have an older format to counteract the simple text and cover various topics They include activities visuals and assessment sheets as well as teacher pages Cambridge IGCSETM Biology Teacher's

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