

Engineering Chemistry Questions and Answers – Chromatography

This set of Engineering Chemistry Multiple Choice Questions & Answers (MCQs) focuses on "Chromatography".

1. Chromatography is the process for identification, purification and separation of components of a mixture on the basis of their boiling point.
- a) Difference in their affinity for mobile and stationary phase solubility
b) Difference in their boiling point
c) Difference in their affinity for mobile and stationary phase solubility
d) Difference in their boiling point

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Answer: c
Explanation: Chromatography is the process for identification, purification and separation of components of a mixture on the basis of difference in their affinity for mobile and stationary phases. The mixture is dissolved in a fluid called the mobile phase, which carries it through a structure holding another material called the stationary phase.

2. Chromatography involves two mutually immiscible phases.
- a) Immiscible phases
b) Miscible phases
c) Soluble phases
d) None of the mentioned

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Answer: a
Explanation: Chromatography involves two mutually immiscible phases (stationary and mobile phases). The mixture is dissolved in a fluid called the mobile phase, which carries it through a structure holding another material called the stationary phase.

3. There is no need of support if packed into a column when the stationary phase is
- a) Gas
b) Liquid
c) Solid
d) None of the mentioned

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Answer: c
Explanation: If the stationary phase is solid, there is no need of support if packed into a column. In solid phase, all the molecules are already in a closed packing.

4. Which of the following is not stationary phase?
- a) Liquid-liquid chromatography
b) Gas-liquid chromatography
c) Gas-solid chromatography
d) Solid-solid chromatography

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Answer: d
Explanation: Solid-solid chromatography is not stationary phase because solid-solid phase cannot provide any fluidity.

5. Liquid chromatography is a technique for separating

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