

Balancing Chemical Equations Worksheet 2

26. $\text{Mg} + \text{Cl}_2 \rightarrow \text{MgCl}_2$
27. $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$
28. $\text{K} + \text{O}_2 \rightarrow \text{K}_2\text{O}$
29. $\text{Cl}_2 + \text{F}_2 \rightarrow \text{ClF}_2$
30. $\text{SiO}_2 + \text{C} \rightarrow \text{Si} + \text{CO}$
31. $\text{NaHCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{CO}_2 + \text{H}_2\text{O}$
32. $\text{ZnS} + \text{O}_2 \rightarrow \text{ZnO} + \text{SO}_2$
33. $\text{SO}_2 + \text{H}_2\text{S} \rightarrow \text{S} + \text{H}_2\text{O}$
34. $\text{Ba} + \text{O}_2 \rightarrow \text{BaO}$
35. $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
36. $\text{SO}_2 + \text{O}_2 \rightarrow \text{SO}_3$
37. $\text{KIO}_3 \rightarrow \text{KI} + \text{O}_2$
38. $\text{NOCl} \rightarrow \text{NO} + \text{Cl}_2$
39. $\text{COF}_2 \rightarrow \text{CO}_2 + \text{CF}_4$
40. $\text{Fe}_2\text{O}_3 + \text{Al} \rightarrow \text{Al}_2\text{O}_3 + \text{Fe}$
41. $\text{P}_4 + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$
42. $\text{Na}_3\text{N} \rightarrow \text{Na} + \text{N}_2$
43. $\text{P}_4 + \text{Cl}_2 \rightarrow \text{PCl}_3$
44. $\text{CrO}_3 \rightarrow \text{Cr}_2\text{O}_3 + \text{O}_2$
45. $\text{P}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{PO}_4$
46. $\text{NaH} + \text{BF}_3 \rightarrow \text{B}_2\text{H}_6 + \text{NaF}$
47. $\text{P}_4 + \text{O}_2 \rightarrow \text{P}_4\text{O}_{10}$
48. $\text{BN} + 3\text{IF} \rightarrow \text{NI}_3 + \text{BF}_3$
49. $\text{Fe}_2\text{O}_3 + \text{C} \rightarrow \text{Fe} + \text{CO}_2$
50. $\text{Fe}_3\text{O}_4 + \text{CO} \rightarrow \text{Fe} + \text{CO}_2$

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Mark Jackson



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