

Pre-Lab Questions – Determination of a Chemical Formula

Answer the following questions using a pen in your laboratory notebooks prior to coming to your regularly scheduled lab section. These pre-lab questions are due when you arrive at lab on the week of October 20. Failure to complete the pre-lab assignment will result in your dismissal from that lab session and you will receive a grade of zero on this lab.

1. What is a chemical formula and what do the subscripts represent in a chemical formula? How do the subscripts of a chemical formula change when a chemical equation is balanced?
2. What is a mole in chemistry? How do the subscripts in a chemical formula directly relate to the concept of a mole?
3. Given 1 molecule of acetic acid (CH_3COOH):
 - a. How many carbon atoms do you have?
 - b. How many hydrogen atoms do you have?
 - c. How many oxygen atoms do you have?
4. Given 1 mole of acetic acid (CH_3COOH):
 - a. How many moles of carbon atoms do you have?
 - b. How many moles of hydrogen atoms do you have?
 - c. How many moles of oxygen atoms do you have?
 - d. How many actual carbon atoms do you have?
 - e. How many actual hydrogen atoms do you have?
 - f. How many actual oxygen atoms do you have?
5. What is the mass of one mole of acetic acid?
6. What are cations? What are anions? How does each fit into a chemical formula of an ionic compound?
7. What is a coefficient in a chemical formula? What role do the coefficients play (why are they included?)
8. What is a limiting reagent in a chemical reaction? How can we use the limiting reagent in a chemical reaction to help us determine the proper chemical formula of a compound?

Chemical Formulas Prelab Questions

**Jerry R. Mohrig, Christina Noring
Hammond, Paul F. Schatz**



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Exploring Physical Science in the Laboratory John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts

Green Chemistry Laboratory Manual for General Chemistry Sally A. Henrie,2015-03-18 Green chemistry involves designing novel ways to create and synthesize products and implement processes that will eliminate or greatly reduce negative environmental impacts Providing educational laboratory materials that challenge students with the customary topics found in a general chemistry laboratory manual this lab manual enables students to see how green chemistry principles can be applied to real world issues Following a consistent format each lab experiment includes objectives prelab questions and detailed step by step procedures for performing the experiments Additional questions encourage further research about how green chemistry principles compare with traditional more hazardous experimental methods

Simulations and Student Learning Matthew Schnurr,Anna MacLeod,2021-01-04 The book underlines the value of simulation based education as an approach that fosters authentic engagement and deep learning

Techniques in Organic Chemistry Jerry R. Mohrig,Christina Noring Hammond,Paul F. Schatz,2006 Is the most comprehensive and detailed presentation of lab techniques available for organic chemistry students and the least expensive It combines specific instructions for 3 different kinds kinds of laboratory glassware and offers extensive coverage of spectroscopic techniques and a strong emphasis on safety issues

Practical Organic Chemistry Workbook Devanshi Magoo,Anju Srivastava,Sriparna Dutta,2025-07-11 This book is a unique laboratory workbook which is intended for the use of undergraduate students of various universities pursuing chemistry as the major subject and has been especially tuned as per the latest NEP UGCF syllabus It is also helpful to students pursuing biochemistry and pharmacy at college or university levels Further this workbook provides basic guidance in the essential experimental techniques that an undergraduate student must master to hone the skills required for training in chemical sciences To the best of our knowledge this is the first time that a workbook for NEP UGCF syllabus is being proposed for helping students acquire prior deep knowledge pertaining to a particular exercise objective and enthuse them by stimulating their scientific thinking through various pre and post lab questions formulated The current pedagogy demands a reformation in the conduct of the practicals under the new syllabi scheme therefore it is anticipated that this manual would be beneficial for the students as well as teachers in rendering the best Besides the book encompasses the fundamentals of all the techniques that would equip the practitioners with specific

skills especially on various purification techniques such as recrystallization and chromatography without which one cannot think about proceeding with any synthesis in any of the chemical pharmaceutical textile industries oil and petroleum industries and more

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

Instructors Manual to Lab Manual Ralph Petrucci, William Harwood, Geoffrey Herring, 2001 *Laboratory Experiments for General, Organic & Biochemistry* Frederick A. Bettelheim, Joseph Marvin Landesberg, 1997 *Ask the Science Expert* Pam Walker, 2001-01-15 Students become the science experts themselves as they read through the advice column format of each problem then develop their own plan for research that will solve the problem and answer the questions They then organize their data and respond to the questions probed in each simulation Includes problems in both life and physical sciences and supports the National Science Education Standards

Chemistry Karen Timberlake, 1999 Suitable for one or two term lab courses covering general organic and biological chemistry this new edition written by Karen Timberlake features many improvements to the insightful experiments that have made it the leading lab manual Each experiment encourages critical thinking with laboratory goals discussion of related concepts clear instructions new pre lab questions and comprehensive report pages Forty one experiments illustrate the basic principles of chemistry

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Carolina Science and Math Carolina Biological Supply Company, 2003 **Experiments in General Chemistry** Kenneth W. Whitten, Carl B. Bishop, Kenneth D. Gailey, 1992 The Software Encyclopedia 2000 Bowker Editorial Staff, 2000-05

Chemistry in the Laboratory James M. Postma, Julian L. Roberts, J. Leland Hollenberg, 2009-08-07 For nearly 40 years Chemistry in the Laboratory has been meeting the needs of teachers and students This new edition builds on that legacy while addressing cutting edge trends in the chemistry laboratory including forensic chemistry and environmental and green chemistry As always the new edition of Chemistry in the Laboratory offers precise easy to follow instructions helpful illustrations and an emphasis throughout on laboratory safety Again throughout a Consider This feature encourages users to

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