

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____.
 - (a) how many atoms of Fe are in one formula unit
 - (b) how many sulfate ions can be attached to the iron atom
 - (c) the charge on each Fe ion
 - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to _____.
 - (a) change the number of moles represented by the formula
 - (b) change the charges on the other ions in the compound
 - (c) change the formula so that it no longer represents the compound it previously represented
 - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.

<u>4 elements</u>	a. How many elements make up this compound?
<u>6 oxygen atoms</u>	b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
<u>21 atoms</u>	c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
<u>4.2×10^{24} atoms</u>	d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
- 4.** How many atoms are present in each of these formula units?

<u>11 atoms</u>	a. $Ca(HCO_3)_2$
<u>45 atoms</u>	b. $C_{12}H_{22}O_{11}$
<u>10 atoms</u>	c. $Fe(ClO_2)_3$
<u>9 atoms</u>	d. $Fe(ClO_3)_2$
- 5.**

<u>N_2O_5</u>	a. What is the formula for the compound dinitrogen pentoxide?
<u>carbon(IV) sulfide</u>	b. What is the Stock name for the covalent compound CS_2 ?
<u>H_2SO_3</u>	c. What is the formula for sulfurous acid?
<u>phosphoric acid</u>	d. What is the name for the acid H_3PO_4 ?

Chemical Names And Formulas Chapter 9 Practice Problems

G. William Daub, William S. Seese



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Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students *Understanding Chemistry* Fred M. Dewey, 1994 *Basic Chemical Principles* Edward I. Peters, 1988

Addison-Wesley Chemistry, 1990 *General Organic and Biological Chemistry* Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features

Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed Peter P. Berlow, 1978

Exercises in General Chemistry Horace Grove Deming, Saul Bryan Arenson, 1924 **General, Organic, and**

Biochemistry Media Update Ira Blei, George Odian, 2008-12-19 Blei and Odian's text gives students the tools they need to develop a working understanding of chemical principles rather than just asking them to memorize facts Now available in a new media enhanced version complete with its own online course space learning environment ChemPortal Blei Odian is better suited than ever to meet the needs of the students taking this course The Media Update version of Blei Odian includes references to dynamic interactive tutorials which provide a step by step walkthrough of concepts and problem solving skills as well as answer specific feedback and practice problems We recognize that all introductory courses are not alike For that

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structure prior to treating the various categories of chemical reactions The organization is designed to get students into chemistry quickly yet methodically The classification of matter begins in Chapter 2 before the chapter on measurement and the coverage of bonding appears in Chapter 8 The elementary concepts of chemistry are presented with an emphasis on mathematical calculations and problem solving strategies Essentials of Chemistry Ralph A. Burns, John William Hill, 1995 The second edition of this chemistry textbook that uses practice examples and applications relating chemistry to our lives and the environment Basic Chemistry G. William Daub, William S. Seese, 1996 Written in a style and language that users without science backgrounds can understand This best selling introduction to the basic principles of chemistry draws on the reader's own experiences through analogies and cartoons to learn difficult concepts The clear systematic thinking approach to problem solving has also been highly praised by reviewers and users alike Countdown sections in each chapter consisting of five review questions keyed to previous material provide readers with a basis for material introduced in the new chapter Study exercises found immediately after new topics are introduced reinforce chapter problem material You and Chemistry marginal application icon relates chemistry to the real world End of chapter essays entitled Elements and Compounds relate the applications of specific elements or compounds to the reader's life **Robinson Chemistry Study Guide** Robinson, 1992 Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson Norman E. Griswold, Henry Fuller Holtzclaw, 1988 **Prentice Hall Chemistry**, 2000

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