

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Class period: \_\_\_\_\_

Unit 8: Chemical Names and Formulas  
Test Review **ANSWER KEY**

Sample questions/Important Topics

1. Please fill in the best answer!

- a) The Periodic Table contains all the elements, arranged according to similarities in their properties.
- b) The nonmetals are on the right hand side of the "staircase"
- c) The metals are on the left hand side of the "staircase"
- d) Although hydrogen is on the left hand side of the "staircase" on the periodic table, it is a non-metal.
- e) The two types of compounds are ionic and covalent.
- f) Ionic compounds are formed when atoms transfer electrons.
- g) Atoms that gain electrons get a negative charge.
- h) Metals tend to form ions with a positive charge.
- i) Crystals are huge conglomerations of + and - ions in a 3-dimensional, repeating pattern.
- j) Molecular compounds are formed when atoms share electrons.
- k) A molecule is an electrically neutral group of atoms that act together as a unit.
- l) Molecular compounds are formed when nonmetals bond together.
- m) Of the two types of compounds, ionic compounds have a much higher melting point.
- n) Ionic compounds are good conductors if they are dissolved in water, (aqueous state).
- o) A(n) chemical formula indicates the makeup of a compound.
- p) A(n) empirical formula tells the smallest whole number ratio of + to - ions within an ionic compound.
- q) A(n) molecular formula tells the exact makeup of 1 molecule of a molecular compound.

# Chemical Names And Formulas Vocabulary Review Answers

**National Cancer Institute (U.S.)**



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