BALANCING CHEMICAL EQUATIONS

Name GROUP

Rewrite and balance the equations below.

1. N, + H, - NH, Nz + 3H2 - 2NH3

2. KClO₃ → KCl + O₂ 2 KClO₃ → 2 KCl + 3 O₂

3. NaCI + F2 - NaF + CI, 2NaCO + F2 -> ZNaF + Cl2

4. H, +0, - HO 2H2+ 02 -> 2H20

5. AgNO₃ + MgCl₂ → AgCl + Mg(NO₃)₂ 2AgNO₃ + MgCl₂ → ZAgOl+

6. AIBr3 + K2SO4 → KBr + AL(SO4)3 2 ALBr3+ 3 K2SO4 → 6 KBr + Al2(SO4)3

7. CH, + O2 + CO2 + HO CH4+ 202 - CO2+ 2420

8. C3H8 + O2 → CO2 + H2O C3H8+ 5O2 → 3CO2+4H2O

9. C₈H₁₈ + O₂ → CO₂ + H₂O C₈H₁₈ + = O₂ → 8CO₂ + 9H₂O

10. FOCI, + NOOH → FO(OH), + NOCI_ FECEX + 3 NOOH → FECOH), +3NOCE

11. P + O2 - P2O6 4P + 502 -> 2P2OF

12. No + H20 → NOOH + H2 2Na+ 2 H20 → 2NaOH + H2

13. Ago - Ag + O2 2Ag20 -> 4Ag + O2

14. $S_8 + O_2 \rightarrow SO_3$ $S_8 + 12.0 = 350_3$

15. CO2 + H20 → C6H12O6 + O2 6CO2 + 6H2O → C6H12O6+ 6O2

16. K + MgBr₂ → KBr + Mg 2 K + MgBr, → 2 KBr + Mg

17. HCI + CaCO, → CaCl, + H2O + CO, 2HCI + CaCO, → Cacl+ H1O + COL

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Exp(Noncons)Mod Chem 90 Tzimopoulo,1990 Lab Experiments Modern Chemistry Harold Clark Metcalfe, 1982 Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Chemical Reactions and Their Equations** Ingo Waldemar Dagobert Hackh, 1921 **Differential Equations with Computer Lab Experiments** Dennis G. Zill, 1995 Includes answers index Authentic Practice Chemistry David G. White, 2000 **Holt Chemistry** Modeling of Chemical Reactions R.W. Carr, 2007 Modeling of Chemical Reactions covers Ralph Thomas Myers, 2004 detailed chemical kinetics models for chemical reactions Including a comprehensive treatment of pressure dependent reactions which are frequently not incorporated into detailed chemical kinetic models and the use of modern computational quantum chemistry which has recently become an extraordinarily useful component of the reaction kinetics toolkit It is intended both for those who need to model complex chemical reaction processes but have little background in the area and those who are already have experience and would benefit from having a wide range of useful material gathered in one volume The range of subject matter is wider than that found in many previous treatments of this subject The technical level of the material is also guite wide so that non experts can gain a grasp of fundamentals and experts also can find the book Chemical Reactions Science Learning Guide NewPath Learning, 2014-03-01 The Chemical Reactions Student useful Learning Guide includes self directed readings easy to follow illustrated explanations guiding guestions inquiry based activities a lab investigation key vocabulary review and assessment review questions along with a post test It covers the following standards aligned concepts Changes of Matter Chemical Reactions Formulas Balancing Equations Types of Chemical Reactions 1 Types of Chemical Reactions 2 Energy in Chemical Reactions Evidence of Chemical Reactions and Chemical Reaction Rates Catalysts Aligned to Next Generation Science Standards NGSS and other state standards

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