

**CHEMICAL REACTIONS TEST (2013-2014)****Physical Science (Standards 1.4, 3.8, 5C)**

1. A chemical reaction is a process in which one or more substances \_\_\_\_\_, resulting in making one or more new substances.
- chemically
  - physically
  - temporarily
  - permanently
2. Which of the following statements is NOT true about chemical reactions?
- A chemical reaction produces new substances with different properties.
  - Chemical reactions result in products that have characteristics that are not the same as the reactants.
  - During chemical reactions, chemical bonds break and/or new bonds form.
  - When a chemical reaction releases energy, it is an example of an endothermic reaction.
3. Chemical reactions always involve chemical changes. Which of the following best describes the final stage of a chemical reaction?
- change in mass, color change, odor change, temperature change
  - odor change, shape change, gas produced, change in the number of atoms
  - gas produced, odor change, temperature change, change in the number of atoms
  - gas produced, color change, temperature change, solid produced
4. Magnesium is a solid gray reactive metal. Hydrochloric acid is a clear liquid with a pH of zero. When Magnesium reacts with Hydrochloric acid, it forms Hydrogen, Magnesium Chloride, and gets hot. Which of the following is not a sign of a chemical change for this reaction?
- Gas was produced
  - Change in the total mass of products
  - Change in temperature
  - New substance formed
5. When a chemical reaction occurs, chemical bonds are broken and new chemical bonds are formed. When this occurs, the products that are made will have physical and chemical properties that are \_\_\_\_\_ reactants.
- the same as
  - similar to
  - opposite of
  - different from
6. When a chemical reaction produces heat, it is called an \_\_\_\_\_ reaction. However, when a chemical reaction absorbs heat, it is called an \_\_\_\_\_ reaction.
- exothermic, endothermic
  - endothermic, exothermic
  - exothermic, exothermic
  - endothermic, exothermic

Sodium [Na] is a soft, extremely malleable metal with a silver-white color. Chlorine [Cl] is a poisonous greenish-yellow gas. When Sodium combines with Chlorine, it reacts to form Sodium

# Chemical Reactions Test B Answer Key

**Mocktime Publication**



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