# Balancing Equations Worksheet - Solutions

2) 
$$_{6}$$
 K +  $_{1}$  B<sub>2</sub>O<sub>3</sub>  $\rightarrow$  \_3 K<sub>2</sub>O + \_2 B

4) 
$$_10$$
\_ Na +  $_2$ \_ NaNO<sub>3</sub>  $\rightarrow$  \_6\_ Na<sub>2</sub>O + \_1\_ N<sub>2</sub>

5) 
$$_4$$
\_ C + \_1\_ S<sub>8</sub>  $\rightarrow$  \_4\_ CS<sub>2</sub>

6) 
$$_4$$
\_ Na +  $_1$ \_ O<sub>2</sub>  $\rightarrow$  \_2\_ Na<sub>2</sub>O

7) 
$$_2$$
\_  $N_2$  +  $_5$ \_  $0_2$   $\rightarrow$   $_2$ \_  $N_20_5$ 

8) 
$$_2$$
\_  $_3$ PO<sub>4</sub> +  $_3$ \_  $_3$ Mg(OH)<sub>2</sub>  $\rightarrow$   $_1$ \_ Mg<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> +  $_6$ \_  $_4$ PO

9) \_2\_ NaOH + \_1\_ 
$$H_2CO_3 \rightarrow _1_ Na_2CO_3 + _2_ H_2O$$

10) \_1\_ KOH + \_1\_ HBr 
$$\rightarrow$$
 \_1\_ KBr + \_1\_ H<sub>2</sub>O

11) \_4\_ Na + \_1\_ 
$$O_2 \rightarrow _2$$
 Na<sub>2</sub>O

12) \_2\_ Al (OH)<sub>3</sub> + \_3\_ H<sub>2</sub>CO<sub>3</sub> 
$$\rightarrow$$
 \_1\_ Al<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub> + \_6\_ H<sub>2</sub>O

13) 
$$_16$$
  $_16$ 

14) 
$$_{6}$$
\_ Cs +  $_{1}$ \_ N<sub>2</sub>  $\rightarrow$   $_{2}$ \_ Cs<sub>3</sub>N

# **Chemical Reactions Ws 1 Balancing Equations Answers**

Ralph H. Petrucci

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