

# Answer Sheet for Sample Problems for Chemistry Level 1 Final Exam 2013-2014 Study Guide

## Electrons in Atoms – Chapter 5

1. What is the frequency of green light, which has a wavelength of  $4.90 \times 10^{-7} \text{ m}$ ?

$$c = \lambda \nu \quad \lambda = 4.90 \times 10^{-7} \text{ m}, c = 3.00 \times 10^8 \text{ m/s} \quad \nu = ? \text{ Hz} \quad \nu = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{4.90 \times 10^{-7} \text{ m}}$$

$6.12 \times 10^{14} \text{ Hz}$

2. An x-ray has a wavelength of  $1.15 \times 10^{-10} \text{ m}$ . What is its frequency?

$$c = \lambda \nu \quad \lambda = 1.15 \times 10^{-10} \text{ m}, c = 3.00 \times 10^8 \text{ m/s} \quad \nu = ? \text{ Hz} \quad \nu = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{1.15 \times 10^{-10} \text{ m}}$$

$2.61 \times 10^{18} \text{ Hz}$

3. Which type of electromagnetic radiation has the most energy: Microwaves, X-Rays, or Visible light? Why?

Ans: X-Rays, b/c they have the shortest wavelength and highest frequency of the types listed and frequency and energy are directly related.

4. What is the energy of a  $9.50 \times 10^{13} \text{ Hz}$  radiation?

$$E = h\nu \quad \nu = 9.50 \times 10^{13} \text{ Hz}, h = 6.626 \times 10^{-34} \text{ J s} \quad E = ? \text{ J}$$

$$E = h\nu = (6.626 \times 10^{-34} \text{ J s}) (9.50 \times 10^{13} \text{ s}^{-1}) = \underline{6.29 \times 10^{-19} \text{ J}}$$

5. The element with the electron configuration of  $[\text{Kr}]5s^2 4d^{10} 5p^1$  is indium (In)

6. Barium (Ba)

7. Write out the electron configurations for the following elements: (For the exam make sure you know how to do both the complete and the noble gas short cut for electron configurations)

a. Tc                      Ans:  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^{10} 4p^6 5s^2 4d^5$  or  $[\text{Kr}]5s^2 4d^5$

b. Aluminum            Ans:  $1s^2 2s^2 2p^6 3s^2 3p^1$  or  $[\text{Ne}] 3s^2 3p^1$

8. Why are Cr and Cu exceptions to writing the electron configurations of elements?

They are exceptions because instead of following the pattern of filling up the d orbitals Cr actually has a half filled s and d orbitals and Cu has a half filled s and completely filled d orbitals. The reason why this occurs is because the electrons in the d-sublevel are more stable with this electron configuration than the expected configuration.

9. a. Ca

b. As

c. Sn

a. 2     Ca.

b. 5     As.

c. 4     Sn.

## Periodic Table and Periodic Trends – Chapter 6

1. Compare & contrast how Mendeleev's periodic table was organized to how Moseley organized it.

# Chemistry 1 2nd Semester Review Answers

**N. H. Morison**



## Chemistry 1 2nd Semester Review Answers:

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que confluyen en la

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of food intolerances these dose dependent yet non immunological reactions require gastroenterological or

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web jan 21 2022 people who have fructose intolerance should limit high fructose foods such as juices apples grapes  
watermelon asparagus peas and zucchini some lower fructose foods such as bananas blueberries strawberries carrots

avocados green beans and lettuce may be tolerated in limited quantities with meals

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web treatment symptoms can be reduced with a low fructose diet following diagnosis it may be helpful for a short period to eat foods low in fructose and avoid sugar alcohols then individual tolerance of fructose is determined in a test phase

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web results intolerance prevalence across all fgids was 60 to fructose 51 to lactose and 33 to both malabsorption occurred in 45 32 and 16 respectively there were no differences in intolerance or malabsorption prevalence between fgid subgroups fgid symptoms correlated with symptoms evoked during testing  $r = 0.35$   $p = 0.0061$

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web a variety of diseases can result in the malabsorption of sugars including lactose intolerance congenital glucose galactose malabsorption and congenital sucrase isomaltase deficiency resulting in the malabsorption of lactose galactose and maltose respectively reference ma long and chen 10 reference burke 11 however this

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usually harmless but its symptoms can be uncomfortable

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web fructose intolerance may refer to fructose malabsorption a digestive disorder of the small intestine in which the fructose carrier in enterocytes is deficient hereditary fructose intolerance a hereditary condition caused by a deficiency of liver enzymes that metabolise fructose this disambiguation page lists articles associated with the

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web fructose and lactose intolerances are common in fgid and associated with increased non gi symptoms but not with specific fgid subtypes symptoms experienced during breath testing but not malabsorption correlate with fgid symptoms effective symptom relief with dietary adaptation is not associated with malabsorption

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