

CHEM 103 Lab Exam 4 (100 out of 100) Questions and Verified Answers (Portage Learning)

Question 1

10 / 10 pts

1. (3 points) What does it mean that the heat of a reaction is an extensive property?

2. (3 points) Does the following reaction give off heat or absorb heat? Explain briefly.



3. (4 points) In one experiment in the video, room temperature water was mixed with water at about 38 degrees. To what temperature (about) did the mixture of the two water samples come to after setting for a few minutes?

1. The heat depends on the amount of substance. If the amount of substance is doubled, then the heat doubles. Other properties, such as density, are intensive, which means the amount of substance does not matter.

2. The (+) value indicates this reaction absorbs heat.

3. The mixture reached 30°C, but settled to 29.8°C.

Question 2

3 / 10 pts

[Scientific Calculator](#)

Answer the following three questions:

1. (3 points) A student conducted an experiment where she mixed 75 g of tap water (25.0°C) with 75g of warm water(35.0°C). The final temperature of the mixture was 28.0°

How much heat was gained by the tap water if the specific heat of water is 4.18 J/K-g?

Chemistry 103 Lab Manual Answers 11ed

James E. Brady



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Index Medicus ,1916 *Microscale Organic Laboratory* Dana W. Mayo,Ronald M. Pike,David C. Forbes,2023-02-07
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Table of Contents Chemistry 103 Lab Manual Answers 11ed

1. Understanding the eBook Chemistry 103 Lab Manual Answers 11ed
 - The Rise of Digital Reading Chemistry 103 Lab Manual Answers 11ed
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 103 Lab Manual Answers 11ed
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 103 Lab Manual Answers 11ed
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 103 Lab Manual Answers 11ed
 - Personalized Recommendations
 - Chemistry 103 Lab Manual Answers 11ed User Reviews and Ratings
 - Chemistry 103 Lab Manual Answers 11ed and Bestseller Lists
5. Accessing Chemistry 103 Lab Manual Answers 11ed Free and Paid eBooks
 - Chemistry 103 Lab Manual Answers 11ed Public Domain eBooks
 - Chemistry 103 Lab Manual Answers 11ed eBook Subscription Services
 - Chemistry 103 Lab Manual Answers 11ed Budget-Friendly Options

6. Navigating Chemistry 103 Lab Manual Answers 11ed eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 103 Lab Manual Answers 11ed Compatibility with Devices
 - Chemistry 103 Lab Manual Answers 11ed Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 103 Lab Manual Answers 11ed
 - Highlighting and Note-Taking Chemistry 103 Lab Manual Answers 11ed
 - Interactive Elements Chemistry 103 Lab Manual Answers 11ed
8. Staying Engaged with Chemistry 103 Lab Manual Answers 11ed
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 103 Lab Manual Answers 11ed
9. Balancing eBooks and Physical Books Chemistry 103 Lab Manual Answers 11ed
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 103 Lab Manual Answers 11ed
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 103 Lab Manual Answers 11ed
 - Setting Reading Goals Chemistry 103 Lab Manual Answers 11ed
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 103 Lab Manual Answers 11ed
 - Fact-Checking eBook Content of Chemistry 103 Lab Manual Answers 11ed
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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