

### Exam 1 – version 1

1. How is the substance carbon monoxide classified? (Hint: It has two elements that have a fixed relative composition. This substance cannot be broken down by physical means.)
- a) Element    **b) Compound**    c) Heterogeneous mixture    d) Homogeneous mixture
5. A men's regulation basketball has a volume of  $43.4 \text{ in}^3$ . What is this volume in cubic centimeters,  $\text{cm}^3$ ?
- a)  $7.11 \times 10^2 \text{ cm}^3$     b)  $1.10 \times 10^2 \text{ cm}^3$     c)  $2.65 \times 10^3 \text{ cm}^3$   
d)  $8.71 \times 10^2 \text{ cm}^3$     e)  $1.34 \times 10^2 \text{ cm}^3$
9. A scientist measured the density of water three times and compared these results with the accepted value which is  $1.000 \text{ g/mL}$ . The following results were obtained:
- Experiment #1:  $0.904 \text{ g/mL}$   
Experiment #2:  $0.902 \text{ g/mL}$   
Experiment #3:  $0.906 \text{ g/mL}$

How would these results be characterized?

- a) high accuracy and high precision  
b) high accuracy and low precision  
**c) low accuracy and high precision**  
d) low accuracy and low precision

How to convert from °C to F and vice versa

Review names of elements

Locate metals/non-metals etc. on periodic table

18. Which of the following elements would be expected to have the most similar chemical and physical properties?
- a) **Magnesium and Calcium**    b) Silver and Iron    c) Oxygen and Nitrogen  
d) Phosphorous and Boron    e) Neon and Fluorine

Review A=mass#, Z=atomic #, q=charge, electrons

23. What is the systematic name for the compound,  $\text{ZnCl}_2$ ?
- a) Zinc Chloride    b) Zinc(II) chloride    c) Zinc dichloride  
d) Zinc chlorine    e) Zinc Chlorine

names of polyatomic ions, naming inorganic compounds

write the metric prefixes

# Chemistry 1150 Final

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