Formula Sheet for Chemistry 1000 H

$E = h\nu$	$B.O. = \frac{1}{2}(n_b - n_a)$	$(\Delta x)(\Delta p_x) \ge \frac{h}{4\pi}$
$\lambda = \frac{h}{\nu(nu)}$	$c = \nu \lambda$	$u = \sqrt{\frac{3RT}{M_m}}$
Vαn	$V\alpha \frac{1}{P}$	VαT
$\chi_A = \frac{n_A}{n_{Total}} = \frac{P_A}{P_{Total}}$	$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$	$\frac{Rate_{_A}}{Rate_{_B}} = \frac{\sqrt{M_{_B}}}{\sqrt{M_{_A}}} = \sqrt{\frac{M_{_B}}{M_{_A}}}$
$(P + \frac{n^2a}{V^2})(V - nb) = nRT$	$pH = pK_o + \log \frac{[A^*]}{[HA]}$	PV = nRT
$pH = -\log(\{H^+\})$	$pK_a = -\log(K_a)$	$K_w = K_a K_b$
$F = \frac{q_1 q_2}{r^2}$	$\ln P_{\text{tip}} = -\left(\frac{\Delta_{\text{tip}} H^{+}}{R}\right) \frac{1}{T} + \ln \beta$	$\ln\left(\frac{P_2}{P_1}\right) = -\left(\frac{\Delta_{mp}H^+}{R}\right)\left[\frac{1}{T_2} - \frac{1}{T_1}\right]$
$\Delta G_{\rm mix} = \Delta H_{\rm mix} - T \Delta S_{\rm mix}$	$S = k_{H}P$	$P_A = P_A^{\sigma} \chi_A$
$P = P_A^o \chi_A + P_B^o \chi_B$	$P = P_{\scriptscriptstyle B}^{\scriptscriptstyle o}(1-\chi_{\scriptscriptstyle A})$	$\Delta T_b = iK_b m$
$\Delta T_f = iK_f m$	$\Pi = MRT$	$K_w = [H_3O^+][OH^-]$

Physical Constants:

$N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$	$h = 6.626 \times 10^{-34} \text{ J s}$	
$m_e = 9.1094 \times 10^{-31} \text{ kg}$	$c = 2.998 \times 10^8 \mathrm{ms^{-1}}$	
$K_w = 1.00 \times 10^{-14}$	$R_H = 2.179 \times 10^{-18} \text{ J}$	

Conversions:

$$1 \text{ J} = 1 \text{ kg m}^2 \text{s}^{-2}$$
 $K = {}^{\circ}\text{C} + 273.15$
 $1 \text{ atm} = 101.3 \text{ kPa}$ $= 760 \text{ mm Hg}$ $= 760 \text{ torr}$ $= 1.01325 \text{ bar}$
 $1 \text{ bar} = 100 \text{ kPa}$ $= 750 \text{ mm Hg}$ $= 750 \text{ Torr}$ $= 0.98692 \text{ atm}$

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Katrin Zwirglmaier

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