

Name: **Answers**

Date:

Quiz#1 Atomic Structure and the Periodic Table, 1.1-1.5

Total: /28 marks

Part 1: Identify the choice that best completes the statement or answers the question.

1. The person given credit for developing the first modern periodic table is
 - A. Dalton
 - B. Mendeleev**
 - C. Democritus
 - D. Chadwick
 - E. Thomson
2. Two atoms are isotopes if they have
 - A. different atomic numbers
 - B. the same mass number, but different atomic numbers
 - C. the same number of protons and neutrons
 - D. the same number of electrons, but a different number of neutrons
 - E. the same atomic number, but a different mass number**
3. The element found in the periodic table in Group 6 and Period 4 is
 - A. hafnium, Hf
 - B. gadolinium, Gd
 - C. selenium, Se
 - D. chromium, Cr**
 - E. lithium, Li
4. Which of the following matches of group number and common name is incorrect?
 - A. Group 2 - alkaline earth metals
 - B. Group 17 - halogens
 - C. Group 6 - lanthanides**
 - D. Group 18 - noble gases
 - E. Group 1 - alkali metals
5. Elements with atomic numbers from 57 to 70 are classified as which of the following?
 - A. transuranic elements
 - B. lanthanides**
 - C. noble gases
 - D. actinides
 - E. transition metals
6. Element X is a light, very reactive metal that forms an oxide with the formula X_2O . This element reacts with water to release hydrogen. Identify the element.
 - A. magnesium
 - B. sodium**
 - C. calcium
 - D. copper
 - E. silver

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