

7.) Naturally-occurring copper exists as two isotopes: ^{63}Cu and ^{65}Cu , the masses of which are given below. Which of these isotopes, if either, is most abundant?

Isotope	Mass (amu)
^{63}Cu	62.929598
^{65}Cu	64.927730

- a. More information is needed
 b. Neither, they are nearly-equal abundance
 c. ^{63}Cu is more abundant
 d. ^{65}Cu is more abundant

8.) The orbital diagram for the valence shell of a phosphorous atom in its ground state is:

	$\uparrow\downarrow$	$\uparrow\downarrow$	\uparrow	
a.	$\uparrow\downarrow$			3p
	3s			
b.	$\uparrow\downarrow$			3p
	3s	\uparrow	\uparrow	\uparrow
c.	$\uparrow\uparrow$			3p
	3s			
d.		$\uparrow\downarrow$	$\uparrow\downarrow$	\uparrow
	3s			3p

Section II – Compounds, Reactions, Mole Concept, Solutions, and Stoichiometry

9.) How many moles of xenon tetrafluoride correspond to 1.036 g of this compound?

- a. 2.5 mmol
 b. 1.5 mmol
 c. 500. mmol
 d. 5.0 mmol
 e. None of these

10.)

How many iodine atoms are present in 39.0 g of $\text{Ca}(\text{IO}_3)_2$?

- a. 6.002×10^{23}
 b. 3.011×10^{23}
 c. 1.20×10^{24}
 d. 1.20×10^{23}
 e. None of these

Chemistry 1 1st Semester Exam Review Answers

Mary Millhollon, Richard H. Langley



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Chemistry Zumdahl, Steven S. Zumdahl, 2003-07 Contains chapter discussions key term definitions and practice chapter tests

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Fundamentals of Probability Saeed Ghahramani, 2018-09-05 The 4th edition of Ghahramani's book is replete with intriguing historical notes insightful comments and well selected examples exercises that together capture much of the essence of probability Along with its Companion Website the book is suitable as a primary resource for a first course in probability Moreover it has sufficient material for a sequel course introducing stochastic processes and stochastic simulation

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