

Name_____

Date_____

Stoichiometry**DIRECTIONS:** Circle the answer that best completes each statement.1. One mole of $\text{Ca}_3(\text{PO}_4)_2$ contains

- (a) one mole of calcium.
- (b) three moles of calcium.
- (c) four moles of oxygen.
- (d) two moles of phosphorus.

a only

b & c

b & d

b only

2. One mole of ammonia (NH_3) contains

- (a) 6.02×10^{23} atoms of nitrogen.
- (b) 14.01 g of nitrogen.
- (c) 7.0 g of nitrogen.
- (d) 6.02×10^{23} molecules.

a, b & d

a, c & d

a & b

d only

3. Which of the following has the greatest mass?

- (a) 200 molecules of water.
- (b) 100 atoms of Fe.
- (c) 200 molecules of O_2 .
- (d) 200 molecules of CH_4 .

a

b

c

d

4. Which of the choices in Question 3 has the least mass?

a

b

c

d

5. Vitamin C, $\text{C}_6\text{H}_8\text{O}_6$

- (a) has an empirical formula of $\text{C}_3\text{H}_4\text{O}_3$.
- (b) contains 40.00% carbon by mass.
- (c) contains 6 moles of carbon atoms per mole of molecules of Vitamin C.
- (d) has a molecular mass of 176.12.

a & c

a, c & d

a, b & c

all of the above

Chemistry 20 Multiple Choice Questions Stoichiometry

Mary Millhollon, Richard H. Langley



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