

Empirical Formula of Copper Sulfide

Lab Report

Table 1 and 2 combined

	Trial 1	Trial 2
Mass of empty crucible (g)	9.70	
Mass of crucible and copper (g)	9.91	
First heating: Mass of crucible and compound (g)	10.11	
Second heating: Mass of crucible and compound (g)	9.98	
Third heating (only if needed): Mass of crucible and compound (g)		
Mass of product (g)		
Mass of copper (g)		
Mass of sulfur (g)		

Calculations

1. For each trial, determine the number of moles of copper in your compound. Show the calculation below and enter the result in Table 2.

Chemistry 210 Lab Report Formulas Copper Sulfide

Tobias Bleicker



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