

Laboratory 2: General Principles of Chemistry (pp. 15-23)

Post-Lab Questions

1. (a) What are three purposes of using a buffered solution in gel electrophoresis?
(b) What would happen if we had used plain water instead of a buffered solution?

- (a) The three purposes of using a buffered solution in gel electrophoresis are because it provides the necessary ions to conduct electricity, it maintains a stable pH and maintains a stable temperature which keeps the gel from melting.
- (b) If we had used plain water instead of a buffered solution the pH would not have been maintained and the pH of the solutions would have changed and created obstacles in the observations.

2. List TWO ways you can speed up molecule migration through the gel during gel electrophoresis. (HINT: Carefully reread bottom of p. 15 and infer.)

Two ways you can speed up molecule migration through the gel during gel electrophoresis is to increase the voltage which runs more energy through the molecules making them move faster and decreasing the gel concentration to enlarge the pore size allowing the molecules to move through the gel easier and faster.

3. (a) How did the temperature change when different voltages were used to run the gel? (Cite the temperature CHANGE results from Table 2.1 for each different voltage.)
(b) What is the general conclusion you can draw about the relationship between voltage and temperature.

- (a) The temperature changed when different voltages were used to run the gel through the temperature increased more when the voltage increased. However, within the results the data was not completely accurate which causes an inaccurate read in the temperature change between 80mV to 100mV causing the table to read a 3.0°C difference for 80mV a 2.5°C difference for 100mV and a 3.5°C difference for 120mV. So, the overall difference did increase between 80 and 120 mVs but the in-between values do not agree.
- (b) The general conclusion you can draw about the relationship between voltage and temperature is that by increasing the voltage there was a greater difference in the temperature of the solution.

4. (a) How did the pH of your electrophoresis chamber change, if at all, over the course of the 30-minute gel run?
(b) WHY was there little to no pH change for this 30-minute run?

Chemistry 2 Laboratory Manual 2015 Answers

Wolfgang Guggemos



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