

Second Long Quiz in Chemistry 8

1. What is called the building block of matter?
A. Atom B. compound C. molecules D. neutron
2. Which has almost the same mass with the proton?
A. electron B. neutron C. nuclei D. orbital



Answer 3-4 based from the atomic notation above.

3. How many protons does magnesium have?
A. 12 B. 24 C. +2 D. no answer
4. How many electron are there in Magnesium?
A. 12 B. 24 C. +2 D. no answer
5. A neutral atom has the same number of proton and electron. Which atomic notation below is correct showing equal number of electron and proton?
A. Na B. Na C. Na D. Na
6. If Magnesium has an atomic number of 12 and it has a charge of +2. Which subatomic particle is affected?
A. electron B. neutron C. proton D. none
7. Atomic mass is the total number of proton and neutron. Which do you think is the heaviest among the following elements?
A. Calcium B. Carbon C. Nitrogen D. Oxygen
8. Recalling the elements, which do you think has the most number of protons and least number of electrons among the elements below?
I. Helium II. Magnesium III. Oxygen IV. Sodium V. Sulfur
A. II, I B. III, I C. IV, I D. V, I
9. How is the atomic mass computed?
A. sum of electrons and protons
B. sum of protons and neutrons
C. sum of neutrons and electrons
D. sum of electrons, neutron and electrons
10. Which total mass is the smallest?
A. The mass of 2 neutrons
B. The mass of 2 electrons
C. The mass of 1 neutron plus mass of 1 proton
D. The mass of 1 electron plus the mass of 1 proton
11. An atom has an atomic number of 48, and a mass number of 120
How many protons, electrons and neutrons does the atom have?
A. The atom has 72 protons, 48 electrons, and 48 neutrons.
B. The atom has 120 protons, 48 electrons, and 72 neutrons.
C. The atom has 72 protons, 72 electrons, and 48 neutrons.
D. The atom has 48, protons, 48 electrons, and 72 neutrons.
12. Which of the following is an element?
A. amalgam B. coal C. silica D. zirconium

Chemistry 8 1 And 8 2guided Reading Answers

Tobias Bleicker



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Complete Foundation Guide For IIT Jee, Chemistry 8 Satyasree Gupta K, Contains large number of Solved Examples and Practice Questions Answers Hints and Solutions have been provided to boost up the morale and increase the confidence level Self Assessment Sheets have been given at the end of each chapter to help the students to assess and evaluate their understanding of the concepts *Properties of Matter: Physical Changes vs. Chemical Changes Gr. 5-8* George Graybill, 2015-09-01 This is the chapter slice Physical Changes vs Chemical Changes from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives *Properties of Matter: Chemical Changes and Chemical Properties Gr. 5-8* George Graybill, 2015-09-01 This is the chapter slice Chemical Changes and Chemical Properties from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions 3rd Edition** Disha Experts, 2019-10-01 As NTA introduces Numeric Answer Questions in JEE Main Disha launches the Questions the 3rd latest updated edition of New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions This study material is developed for quick revision and practice of the complete syllabus of the JEE Main Exam in a short span of 40 days The book can prove to be the ideal material for class 12 students as they can utilise this book to revise their preparation immediately after the board exams The book contains 27 chapters of class 11 12 and each Chapter contains JEE Main 6 Years at a Glance i e JEE Main 2019 2014 with TOPIC WISE Analysis Detailed Concept Maps covers entire JEE Syllabus for speedy revision IMPORTANT CRITICAL Points of the Chapter for last minute revision TIPS to PROBLEM SOLVING to help students to solve Problems in shortest possible time Exercise 1 CONCEPT BUILDER A Collection of Important Topic wise MCQs to Build Your Concepts

Exercise 2 CONCEPT APPLICATOR A Collection of Quality MCQs that helps sharpens your concept application ability

Exercise 3 Numeric Answer Questions A Collection of Quality Numeric Answer Questions as per the new pattern of JEE

Answer Keys Detailed Solutions of all the Exercises and Past years problems are provided at the end of the chapter

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