Section A

For each question there are four possible answers A, B, C and D. Choose the one you consider to be correct.

Use of the Data Booklet may be appropriate for some questions.

1 Compound X is an organic compound that contains 30.6% carbon, 3.8% hydrogen, 20.4% oxygen and 45.2% chlorine by mass.

What is the empirical formula of X?

A C₂H₃OCI

B C₂H₄OCI

C C₂H₄OC/

D C₄H₃O₂CI₂

2 A sample of propane, C₃H₈, with a mass of 9.61 g is completely combusted in an excess of oxygen under room conditions.

Which volume of carbon dioxide gas is produced?

A 4,89 dm³

B 5.24 dm³

C 14.7 dm³

D 15.7 dm³

3 Which atom has the same number of electrons as an ammonium ion?

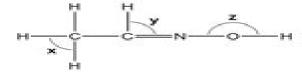
A Mg

B Na

C Ne

D O

4 Ethanal reacts with hydroxylamine, NH₂OH, to form the molecule shown.



What is the order of increasing bond angle in this structure from smallest to largest?

A z. x. y

B y. z. x

C x, z, y

D z, y, x

Chemistry 9701 June 02 Paper 2 Answer

Peter Cann, Peter Hughes

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Table of Contents Chemistry 9701 June 02 Paper 2 Answer

- 1. Understanding the eBook Chemistry 9701 June 02 Paper 2 Answer
 - The Rise of Digital Reading Chemistry 9701 June 02 Paper 2 Answer
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry 9701 June 02 Paper 2 Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 9701 June 02 Paper 2 Answer
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry 9701 June 02 Paper 2 Answer
 - Personalized Recommendations

- Chemistry 9701 June 02 Paper 2 Answer User Reviews and Ratings
- Chemistry 9701 June 02 Paper 2 Answer and Bestseller Lists
- 5. Accessing Chemistry 9701 June 02 Paper 2 Answer Free and Paid eBooks
 - o Chemistry 9701 June 02 Paper 2 Answer Public Domain eBooks
 - Chemistry 9701 June 02 Paper 2 Answer eBook Subscription Services
 - Chemistry 9701 June 02 Paper 2 Answer Budget-Friendly Options
- 6. Navigating Chemistry 9701 June 02 Paper 2 Answer eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Chemistry 9701 June 02 Paper 2 Answer Compatibility with Devices
 - Chemistry 9701 June 02 Paper 2 Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 9701 June 02 Paper 2 Answer
 - Highlighting and Note-Taking Chemistry 9701 June 02 Paper 2 Answer
 - o Interactive Elements Chemistry 9701 June 02 Paper 2 Answer
- 8. Staying Engaged with Chemistry 9701 June 02 Paper 2 Answer
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - o Following Authors and Publishers Chemistry 9701 June 02 Paper 2 Answer
- 9. Balancing eBooks and Physical Books Chemistry 9701 June 02 Paper 2 Answer
 - \circ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 9701 June 02 Paper 2 Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 9701 June 02 Paper 2 Answer
 - Setting Reading Goals Chemistry 9701 June 02 Paper 2 Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 9701 June 02 Paper 2 Answer
 - Fact-Checking eBook Content of Chemistry 9701 June 02 Paper 2 Answer

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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