

# Determining Salt Concentration Using Two Methods

Name \_\_\_\_\_

Instructor \_\_\_\_\_ Section \_\_\_\_\_ Partner \_\_\_\_\_

(20 pts for the first table; 12 pts for the second table)

**Data****Density**

Tables of mass and density values for the salt solutions.

Salt Solution Percentage (use the values on the stock bottles)	Mass of the empty beaker	Mass of 10.00mL solution with the beaker	Mass of 10.00mL of the solution alone	Density of the solution in g/mL
2.000 %	77.161 g	87.527 g		
4.000 %	77.481 g	87.614 g		
6.000 %	77.523 g	87.832 g		
8.000 %	77.506 g	87.710 g		
10.000 %	77.312 g	87.722 g		

Unknown Solution	Mass of the empty beaker	Mass of 10.00mL solution with the beaker	Mass of 10.00mL of the solution alone	Density of the unknown solution in g/mL
A	77.623 g	87.633 g		
B	77.451 g	87.595 g		
C	77.311 g	87.304 g		

# Chem 1211 Lab Manual Answer Key

**EW Minium**



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*Chemistry 1211, LAB Manual* Jack Duff, Alan Gabrielli, 2014-08-01      **Books in Print Supplement**, 1988      **Chemical Regulation Reporter** Bureau of National Affairs (Arlington, Va.), 1981      **Subject Guide to Books in Print**, 1997  
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## **Chem 1211 Lab Manual Answer Key** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chem 1211 Lab Manual Answer Key**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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