

Chpt. 3: Stoichiometry- Mass, Formulas, & Reactions

1. Air, Life, and Molecules

- a. The Earth formed 4.6 billion years ago was a hot, molten sphere that gradually separated into distinct regions based on differences in
 - i. Density
 - ii. Melting point
- b. Volatile gases that were released reacted to change the composition of Earth's atmosphere & surface
- c. Friedrich Wohler (1800-1882)
 - i. Attempted to synthesize ammonium cyanate
 - ii. Succeeded in making urea "without the intervention of a kidney"
 - iii. Led to modern definition of **organic compounds**- compounds of carbon & hydrogen, often including other elements
 - iv. Organic compounds identical to those found in the living systems **CAN** be made in the laboratory
 - v. Ammonium cyanate and urea have same formula $\rightarrow \text{CH}_4\text{N}_2\text{O}$
 1. Scientists concluded the arrangement of atoms must be different
 2. Defined by # of atoms & arrangement of atoms
 3. **Isomers**- two or more different compounds with the same atomic composition
- d. Chemical Reactions & Earth's Early Atmosphere
 - i. Earth's early crust was torn by the impact of asteroids & widespread volcanic activity
 - ii. Early atmospheric air:
 1. Molecular oxygen (O_2)
 2. Carbon dioxide (CO_2)
 3. Carbon monoxide (CO)
 4. Water vapor (H_2O)
 - iii. **Combination reaction**- two or more substances combine to form one product
 1. **Reactants**- a substance consumed during a chemical reaction
 - a. Sulfur trioxide + water \rightarrow sulfuric acid
 2. **Product**- a substance formed during a chemical reaction
 - a. Sulfur + water \rightarrow sulfuric acid
 - iv. Chemical equations:
 1. Describe the proportions of reactants & products during a

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Stephen Cohen, Richard C. Burns



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Tests and Measurements in High School Instruction Giles Murrell Ruch, George Dinsmore Stoddard, 1927

Reading Tests and Reviews Oscar Krisen Buros, 1968 **Resources in Education**, 1995-04 **Timetable** University of Illinois at Urbana-Champaign, 2002 Clinical Chemistry - E-Book Donna Larson, 2015-12-17 Gain a clear understanding of pathophysiology and lab testing Clinical Chemistry Fundamentals and Laboratory Techniques prepares you for success as a medical lab technician by simplifying complex chemistry concepts and lab essentials including immunoassays molecular diagnostics and quality control A pathophysiologic approach covers diseases that are commonly diagnosed through chemical tests broken down by body system and category such as respiratory gastrointestinal and cardiovascular conditions Written by clinical chemistry educator Donna Larson and a team of expert contributors this full color book is ideal for readers who may have minimal knowledge of chemistry and are learning laboratory science for the first time Full color illustrations and design simplify complex concepts and make learning easier by highlighting important material Case studies help you apply information to real life scenarios Pathophysiology and Analytes section includes information related to diseases or conditions such as a biochemistry review disease mechanisms clinical correlation and laboratory analytes and assays Evolve companion website includes case studies and animations that reinforce what you've learned from the book Laboratory Principles section covers safety quality assurance and other fundamentals of laboratory techniques Review questions at the end of each chapter are tied to the learning objectives helping you review and retain the material Critical thinking questions and discussion questions help you think about and apply key points and concepts Other Aspects of Clinical Chemistry section covers therapeutic drug monitoring toxicology transplantation and emergency preparedness Learning objectives in each chapter help you to remember key points or to analyze and synthesize concepts in clinical chemistry A list of key words is provided at the beginning of each chapter and these are also bolded in the text Chapter summaries consist of bulleted lists and tables highlighting the most important points of each chapter A glossary at the back of the book provides a quick reference to definitions of all clinical chemistry terms **Engineering Record, Building Record and Sanitary Engineer** Henry Coddington Meyer, Charles Frederick Wingate, 1911 **New York Medical Journal, and Philadelphia Medical Journal**, 1915 **Army-Navy-Air Force Register and Defense Times**, 1908 **Industrial Management**, 1920 **The Cosmopolitan**, 1921 **Publishers Directory**, 1989 *International Record of Medicine and General Practice Clinics* Edward Swift Dunster, Frank Pierce Foster, James Bradbridge Hunter, Charles Eucharist de Medicis Sajous, Gregory Stragnell, Henry J. Klaunberg, Félix Martí-Ibáñez, 1915 **Educational Research Bulletin**, 1948 Includes the sections Educational readings and Books to read *Pathways of the Pulp* Stephen Cohen, Richard C. Burns, 1998 Pathways of the Pulp has been a market leader and book of choice in dental education and practice for many editions The 8th edition has been completely revised and updated to bring you the latest techniques and materials developed for root canal work apical

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Industrial Management John R. Dunlap,1920 **Legislature's Full Agenda of Problems** ,1983 **The Right College, 1991** College Research Group of Concord Massac,1990-07 The brand new totally updated edition of the college guide covering every aspect of campus life Complete profiles of the more than 1 500 accredited four year colleges and universities in the United States Canada and Mexico cover everything from admissions to prominent graduates to social events **U.S. News & World Report** ,1986

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