

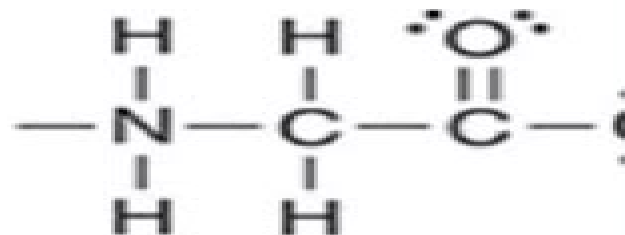
## Organic Chemistry 1 ACS Questions 2024 Final Exam multiple choice with Answers

Which is the correct hybridization state and geometry for the carbon atom in HCN?

- sp, linear
- sp<sup>2</sup>, trigonal planar
- sp<sup>3</sup>, tetrahedral
- None of the above
- sp, linear

The following structure has been drawn without formal charges. Which statement describes the missing formal charge(s)?

- This structure has one positive charge and one negative charge.
- This structure has one positive charge but no negative charges.
- This structure has one negative charge but no positive charges.
- This structure has no formal charges.
- This structure has one positive charge and one negative charge.



The indicated  $\sigma$  bond results from the overlap of which orbitals?

- sp<sup>2</sup>-sp<sup>2</sup>
- sp-sp<sup>3</sup>
- sp-sp<sup>2</sup>
- sp<sup>2</sup>-sp<sup>3</sup>
- sp<sup>2</sup>-sp<sup>3</sup>

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**Guillaume Favre**

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