

CHEM 116 STUDY GUIDE EXAM EVALUATION QUESTIONS WITH ANSWERS 2025/2026

Which one of the following processes produces a decrease in the entropy of the system? - freezing water to form ice

DeltaS is positive for the reaction - $2\text{SO}_3(\text{g}) \rightarrow 2\text{SO}_2(\text{g}) + \text{O}_2(\text{g})$

Which reaction produces a decrease in the entropy of the system? - $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l})$

Which reaction produces an increase in the entropy of the system? - $\text{CO}_2(\text{s}) \rightarrow \text{CO}_2(\text{g})$

Which of the following processes produces a decrease of the entropy of the system? - dissolving oxygen in water

DeltaS is negative for the reaction - $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{SO}_3(\text{g})$

DeltaS is positive for the reaction - $2\text{H}_2\text{O}(\text{g}) \rightarrow 2\text{H}_2(\text{g}) + \text{O}_2(\text{g})$

DeltaS is negative for the reaction - $\text{Mg}(\text{NO}_3)_2(\text{aq}) + 2\text{NaOH}(\text{aq}) \rightarrow \text{Mg}(\text{OH})_2(\text{s}) + 2\text{NaNO}_2(\text{aq})$

Which one of the following correctly indicates the relationship between the entropy of a system and the number of different arrangement, W, in the system? - $S = k \ln W$

The entropy change accompanying any process is given by the equation - $\Delta S = k \ln(W_{\text{final}}/W_{\text{initial}})$

Of the following, the entropy of gaseous ____ is the largest at 25 C and 1 atm... - C_2H_6

Of the following, the entropy of ____ is the largest. - $\text{HI}(\text{g})$

For an isothermal process, the entropy change of the surroundings is given by the equation - $\Delta S = -q_{\text{sys}}/T$

The standard Gibbs free energy of formation of ____ is zero. a) $\text{H}_2\text{O}(\text{l})$ b) $\text{O}(\text{g})$ c) $\text{H}_2(\text{g})$ - C only - $\text{H}_2(\text{g})$

The standard Gibbs free energy of formation of ____ is zero. a) $\text{H}_2\text{O}(\text{l})$ b) $\text{Na}(\text{s})$ c) $\text{H}_2(\text{g})$ - B and C - b) $\text{Na}(\text{s})$ c) $\text{H}_2(\text{g})$

The standard Gibbs free energy of formation of ____ is zero. a) $\text{Al}(\text{s})$ b) $\text{Br}_2(\text{l})$ c) $\text{Hg}(\text{l})$ - All - a) $\text{Al}(\text{s})$ b) $\text{Br}_2(\text{l})$ c) $\text{Hg}(\text{l})$

Chem Solutions Study Guide Answers

Iris Stovall



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review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book *Study Guide and Solutions Manual to accompany Basic Concepts of Chemistry, 9th Edition* Leo J.

Malone,2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone s Basic Concepts of Chemistry

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Organic Chemistry Marye Anne Fox, James K. Whitesell, 2004 Accompanying CD-ROM has been enhanced with updated animated illustrations to accompany the presentations and Chem3D files for helpful structure visualization.

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Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better. The solutions manual section includes detailed answers and explanations to the odd-numbered exercises in the text.

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