

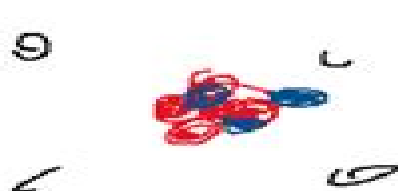
Chemthink guided notes – Ions

Note: There are no “back” buttons on this tutorial so be careful not to click “continue” on the slide until you have all of the information.

At certain times atoms can gain or lose one or more ____electrons_____. When this happens we say that the atom has formed an _ion_____. Ions are just individual atoms or groups of atoms that has a charge. This charge comes from an imbalance of in the number of ____positive_____ protons and ____negative_____ electrons.

Since the electrons don't contribute to the ____mass_____ of the atom, gaining or losing electrons does not change the overall mass of the atom. It does affect it in other ways. Ions can behave very differently than the atoms from which they were formed. As we will see in later units, forming ions is one way that atoms bond together to make new substances.

Draw a picture of the lithium atom and write the particles in the particle inventory



Protons 3

Electrons 3

Neutrons 4 _____

total charge = 0

symbol: Li

What happens when an electron is lost?

symbol?

Draw a picture of the oxygen atom and write the particles in the particle inventory



Protons 8

Electrons 8

Neutrons 8 _____

total charge = 0

symbol: O

What happens when two electrons are gained?

An ion of -2 charge forms

symbol? O^{2-}

Chem Think Lab Answers

**Michael R. Abraham, Michael J.
Pavelich**



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Report Commonwealth Shipping Committee,1911 Organic Chemistry Education Research into Practice Jay Wackerly,Sarah Zingales,Michael Wentzel,Gautam Bhattacharyya,Brett McCollum,2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

Laboratory Exercises for a Brief Course in Chemistry Lyman Churchill Newell,1927 *Inquiries into Chemistry* Michael R. Abraham,Michael J. Pavelich,1999-05-20 The laboratory course should do more than just acquaint the students with fundamental techniques and procedures The laboratory experience should also involve the students in some of the kinds of mental activities a research scientist employs finding patterns in data developing mathematical analyses for them forming hypotheses testing hypotheses debating with colleagues and designing experiments to prove a point For this reason the student tested lab activities in *Inquiries into Chemistry* 3 E have been designed so that students can practice these mental activities while building knowledge of the specific subject area Instructors will enjoy the flexibility this text affords They can select from a comprehensive collection of structured guided inquiry experiments and a corresponding collection of open inquiry experiments depending on their perception as to what would be the most appropriate method of instruction for their students Both approaches were developed to encourage students to think logically and independently to refine their mental models and to allow students to have an experience that more closely reflects what occurs in actual scientific research

Thoroughly illustrated appendices cover safety in the lab common equipment and procedures **The Function of the Laboratory in Secondary Education** Silas Ellsworth Coleman,1900 **Annual Report of the State Department of Health of New York for the Year Ending December 31 ...** New York (State). Department of Health,1910 **An Introduction to the Practice of Organic Chemistry in the Laboratory** Homer Adkins,Samuel Marion McElvain,1925

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National Note-book Sheets for Laboratory Work in Chemistry Arthur Stone Dewing, 1909 *Canadian Chemistry and Metallurgy*, 1927 **Teaching First Year Chemistry** Joseph Otto Frank, 1927 Laboratory Inquiry in Chemistry Richard C. Bauer, Richard Bauer, James P. Birk, Douglas J. Sawyer, 2005 LABORATORY INQUIRY IN CHEMISTRY Second Edition provides a unique set of guided inquiry investigations that focus on constructing knowledge about the conceptual basis of laboratory techniques instead of simply learning techniques By focusing on developing skills for designing experiments solving problems thinking critically and selecting and applying appropriate techniques the authors expose students to a realistic laboratory experience typical of the practicing chemist The Second Edition features six new experiments and is accompanied by a revised and updated Instructor's Manual available online This new edition continues the proven three phase learning cycle exploration of chemical behaviors within the context of the problems posed concept invention the use of data and observations to construct accepted scientific knowledge about the concepts explored in the laboratory investigation and concept application where students apply their conceptual understanding of the investigation at hand by modifying or extending the experiments and write a report that emphasizes conceptual relevance These college and honors level inquiry based experiments correlate well with the recommended experiments outlined by the Advanced Placement Chemistry Development Committee **Sessional Papers** Great Britain. Parliament. House of Commons, 1904 Transactions of the Institution of Chemical Engineers, 1925 *Annual Report of the Regents*, 1893 *Annual Report of the Regents* University of the State of New York, 1893 No 104 117 contain also the Regents bulletins

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