

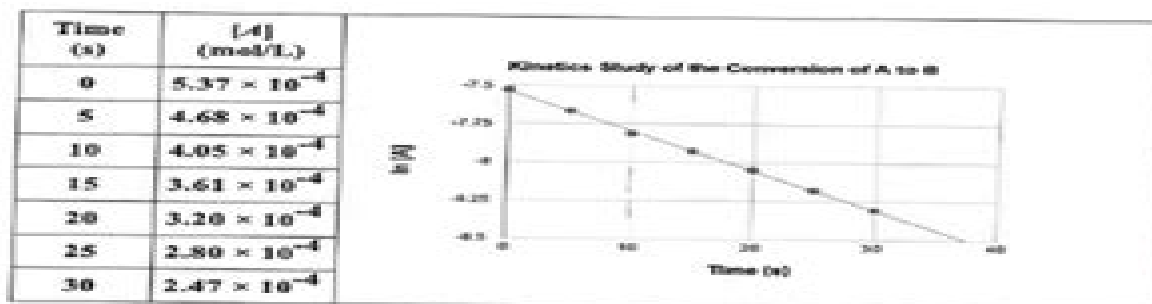
Question 1

Not complete

Marked out of 1.00

A student was studying the decomposition of reactant A to form product B by measuring the concentration of A every five seconds for thirty seconds. The raw data is tabulated below. The student then prepared plots of $[A]$ versus time, $\ln[A]$ versus time, and $1/[A]$ versus time. Only one of these plots revealed a linear relationship; this plot is also shown below. Spreadsheet software was used to generate an equation for the best-fit line for the plot:

$$y = -0.0257x - 7.54$$



Spreadsheet software, although powerful, will not include units when generating the equation for a best-fit line. What should be the units of slope for the best-fit line described above?

Select one:

- ☐ s
☐ mol/L
☐ none
☐ L/mol
☒ s^{-1}
☐ mol

$$\text{slope} = \frac{\text{Rise}}{\text{Run}} = \frac{\Delta y}{\Delta x} \quad \begin{array}{l} \checkmark \text{ No unit} \\ \leftarrow s \end{array} = \frac{1}{s}$$

Chem 102 Lab Manual Answer Key

WJ Hussar



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for 1929 include section Contents noted 1929 1939 called Metallurgical abstracts Jan 1940 Sept 1945 called Engineering
digest Oct 1945 called Materials beginning in 1942 included in the complete index to the periodical *El-Hi Textbooks and*
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