

HONORS CHEM 1st SEMESTER EXAM STUDY GUIDE FALL 2010

STRATEGY: Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a starting point to identify the areas on which you need to spend more study time. For those areas, go back to homework assignments, quizzes, and reviews to practice more problems. I would also recommend going through all of your tests since these questions are only samples and do not include specific examples of how vocabulary and other conceptual information might appear in a multiple-choice or other format. Remember you can access notes and reviews on my website.

IMPORTANT INFORMATION:

- This must be handed in BEFORE taking your final exam.
- If entirely complete, 5 points will be added to your exam grade
- All work must be shown in DETAIL on lined paper

Introduction to Chemistry – Chapter 1

1. Identify safe/unsafe lab practices.
2. List pieces and uses of lab equipment.
3. List the requirements for tables and graphs.
4. What is plotted on the x-axis and on the y-axis?
5. Make a density graph.
6. What is water displacement?

Matter and Change – Chapter 2

7. Classify the following substances as *solid*, *liquid*, *gas*, or *plasma* based on their properties.
 - a. flexible volume, high KE, particles can disperse freely.
 - b. flexible volume, very high KE, particles are charged.
 - c. fixed volume, very low KE, orderly particles.
 - d. fixed volume, low KE, particles can move past each other.
8. Classify the following as element, compound, heterogeneous mixture, or solution.
 - a. graphite (carbon)
 - b. grape juice
 - c. table salt (NaCl)
 - d. pepper
9. Classify the following as *chemical* or *physical changes*.
 - a. cutting wire
 - b. ripening tomato
 - c. apple slices turning brown
 - d. compressing a gas
10. Classify the following properties as *physical* or *chemical*.
 - a. melts at 68.0°C
 - b. corrosive
 - c. reacts violently with water
 - d. decomposes in air
 - e. magnetic
11. Classify the following properties as *intensive* or *extensive*.
 - a. melting point
 - b. volume
 - c. mass
 - d. reactivity
 - e. density

VOCAB: kinetic molecular theory

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J Spring



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Table of Contents Chem Honors Final Review Study Guide

1. Understanding the eBook Chem Honors Final Review Study Guide
 - The Rise of Digital Reading Chem Honors Final Review Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chem Honors Final Review Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem Honors Final Review Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chem Honors Final Review Study Guide
 - Personalized Recommendations
 - Chem Honors Final Review Study Guide User Reviews and Ratings
 - Chem Honors Final Review Study Guide and Bestseller Lists
5. Accessing Chem Honors Final Review Study Guide Free and Paid eBooks
 - Chem Honors Final Review Study Guide Public Domain eBooks
 - Chem Honors Final Review Study Guide eBook Subscription Services
 - Chem Honors Final Review Study Guide Budget-Friendly Options

6. Navigating Chem Honors Final Review Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Chem Honors Final Review Study Guide Compatibility with Devices
 - Chem Honors Final Review Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chem Honors Final Review Study Guide
 - Highlighting and Note-Taking Chem Honors Final Review Study Guide
 - Interactive Elements Chem Honors Final Review Study Guide
8. Staying Engaged with Chem Honors Final Review Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem Honors Final Review Study Guide
9. Balancing eBooks and Physical Books Chem Honors Final Review Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem Honors Final Review Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chem Honors Final Review Study Guide
 - Setting Reading Goals Chem Honors Final Review Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chem Honors Final Review Study Guide
 - Fact-Checking eBook Content of Chem Honors Final Review Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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