

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: \uparrow temp = \uparrow speed

type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

\downarrow
b/c molecules moving faster

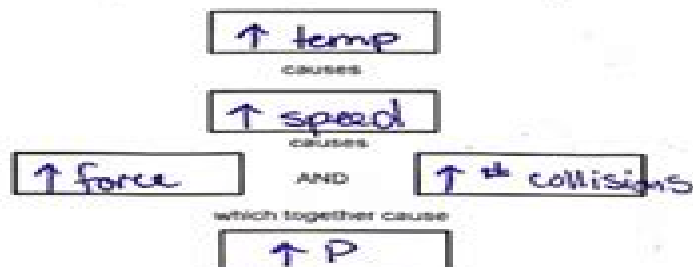
$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

Chem Think Behavior Of Gases Answers

Liying Dong



Chem Think Behavior Of Gases Answers:

Chemistry and Our Universe , In the first of two lectures on the properties of gases review the basic equations that describe their behavior Learn the history of Boyle s law Gay Lussac s law Charles s law and Avogadro s law Then use these four expressions to derive the celebrated ideal gas law **Nonideal Behavior of Gases and Their Mixtures** Abdus

Satter,1963 The Dynamical Theory of Gases (1916) J. H. Jeans,2008-06-01 This scarce antiquarian book is a facsimile reprint of the original Due to its age it may contain imperfections such as marks notations marginalia and flawed pages Because we believe this work is culturally important we have made it available as part of our commitment for protecting preserving and promoting the world s literature in affordable high quality modern editions that are true to the original work

The Laws Of Gases Memoirs Robert Boyle,2023-07-18 First published in 1672 this book is a landmark work in the history of chemistry In it Boyle laid out his theory of the behavior of gases including his famous law that describes the relationship between pressure and volume With its clear explanations and precise experiments this book is a must read for anyone interested in the history of scientific discovery This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant **Gases and "airs", Nature of Air**

and Behavior of Gases. Teacher's Kit ,1974 *The Experimental Study of Gases* Morris William Travers,2025-05-22 The Experimental Study of Gases by Morris William Travers provides a detailed account of the experimental methods used in the study of gases at the turn of the 20th century This comprehensive work is invaluable for students and researchers interested in the history of chemistry and physics as well as the development of laboratory techniques Travers meticulously describes the apparatus procedures and challenges associated with gas research offering insights into the pioneering efforts to understand the properties and behavior of gases The book covers a range of topics including gas preparation purification measurement and analysis It serves as a practical guide for conducting experiments and a historical record of the evolution of experimental science With its clear and precise explanations The Experimental Study of Gases remains a relevant resource for anyone seeking a deeper understanding of the foundations of modern chemistry and physics This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work was reproduced from the original artifact and remains as true to the original work as possible Therefore you will see the original copyright references library stamps as most of these works have been housed in our most important libraries around the world and other notations in the work This work is in the public domain in the United States of America and possibly other

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Phase Claire Vallance,2017-11-23 An Introduction to the Gas Phase is adapted from a set of lecture notes for a core first year lecture course in physical chemistry taught at the University of Oxford The book is intended to give a relatively concise introduction to the gas phase at a level suitable for any undergraduate scientist After defining the gas phase properties of gases such as temperature pressure and volume are discussed The relationships between these properties are explained at a molecular level and simple models are introduced that allow the various gas laws to be derived from first principles Finally the collisional behaviour of gases is used to explain a number of gas phase phenomena such as effusion diffusion and thermal conductivity

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