

Figure 13.4: MO diagrams of 7-, 8-, 9-, and 10-atom cyclic conjugated π systems.

20. Use the MO diagram of an 8-atom ring (above) to fill in the following table.

No. Of Electrons in the π System	2	4	6	8	10	12
Has Radical Character According to MO						
Aromatic (according to Hückel's Rule)						

21. Confirm that you get the same answer to the previous question no matter which cyclic MO diagram you use to fill in the table (as long as there is room for all the electrons).

Term Introduction: Anti-Aromatic Molecules

A molecule is **anti-aromatic** (and highly reactive) if it exhibits all characteristics of aromaticity EXCEPT that it has $4n$ (instead of $4n+2$) π electrons (e.g., 4, 8, 12, etc.).

22. Label each species in Figure 18.3 that is anti-aromatic.
23. Cyclobutadiene can be isolated and studied at temperatures lower than -200°C . At temperatures above -200°C it immediately reacts with itself to form other more stable products. Use molecular orbital theory to explain why **cyclobutadiene** is unstable.



Cyclobutadiene (very unstable)

24. List four things about cyclobutadiene that make it anti-aromatic.

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