

12. Use your answers to the CTQs above to calculate the value of  $K_c$  for  $\text{Mg}(\text{OH})_2$ .
13. Under what circumstances when  $\text{Mg}^{2+}(\text{aq})$  and  $\text{OH}^{-}(\text{aq})$  are present in a solution is the concentration of  $\text{Mg}(\text{OH})_2(\text{s})$  not equal to 40.5 mole/L?

## Information

Notice that for the dissolution of any solid the expression for  $K_c$  will always contain the concentration of the solid in the denominator. Because the concentration of the solid is always constant (at a particular temperature), the concentration of the solid can be combined with  $K_c$  as follows (for  $\text{Mg}(\text{OH})_2$ ):

# Chemactivity 41 Answers

**Michael P. Garoutte**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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