

### Model 2: An Electron Between Two Nuclei.

If an electron is placed halfway between the nuclei, then each nucleus is attracted toward the electron.

Figure 2. Nuclei are attracted to an electron.



The (attractive) Coulombic potential energy is given by the sum of the interactions of nucleus A with the electron and nucleus B with the electron.

$$V = \frac{-2.31 \times 10^{-28} \text{ J} \cdot \text{pm}}{50 \text{ pm}} + \frac{-2.31 \times 10^{-28} \text{ J} \cdot \text{pm}}{50 \text{ pm}} = -9.24 \times 10^{-18} \text{ J (attractive)} \quad (2)$$

The net Coulombic potential energy is the sum of the energy of interaction of the two nuclei (given in equation 1) and the energy of interaction of the two nuclei with the electron (given in equation 2):

$$V = 2.31 \times 10^{-18} \text{ J} + (-9.24 \times 10^{-18} \text{ J}) = -6.93 \times 10^{-18} \text{ J} \quad (\text{attractive})$$

### Critical Thinking Questions

2.
  - a) Why is "50 pm" in the denominator of each term in equation 2?
  - b) Show explicitly how the value for the numerator in both terms of equation 2 was obtained from the Coulombic potential energy expression.
3. Explain why the numerator in equation 1 is positive, but the numerator in equation 2 is negative.

# Chemactivity 32 32 Answers

**Michael P. Garoutte**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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*Prentice Hall Chemistry*, 2000

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