

# **Nuclear Chemistry** (used in activities)

**Model: Models represent the three isotopes of carbon:**



## **Critical Thinking Questions**

1. How many protons are in carbon-12?  $\frac{6}{12}$  How many neutrons are in carbon-12?  $\frac{6}{12}$
2. How many electrons are in carbon-12?  $\frac{6}{12}$  How many neutrons are in carbon-12?  $\frac{6}{12}$
3. How many protons are in carbon-13?  $\frac{6}{13}$  How many neutrons are in carbon-13?  $\frac{7}{13}$
4. How is carbon-14 different from carbon-12? What is different among isotopes?  
(Note: in previous 3 questions)
5. What does the subscript represent in all three models (carbon-12, carbon-13, carbon-14)?

## **Information: Nuclear reactions and binding/decay:**

- A nuclear reaction is a change in the composition of the nucleus of an atom. That is not necessarily accompanied by chemical reaction, and does not depend on what substances are there (except for the).
- There are three types of nuclear reactions: fusion, fission, and radioactive decay. Fusion is the joining of two light nuclei, such as the stars where the sun and fusion. Fission is the splitting of a heavy nucleus, such as the bomb where the atom is split. Radioactive decay is the process of a nucleus changing into a more stable nucleus.
- Some reactions are exothermic, meaning that they release energy. Some reactions are endothermic, meaning that they absorb energy. Some reactions are both, meaning that they release energy in some cases and absorb energy in others.

**Table 1: Types of nuclear reactions and their products**

Type of Reaction	Symbol	Mass Number	Charge	Energy Released/Absorbed	Products	Isotopes Produced
Fusion	$\alpha$ , $\beta$	4	2+	very high	Helium-4	Helium-4 (stable)
Fission	$\alpha$ , $\beta$	4	2+	very high	Helium-4	Helium-4 (stable)
Radioactive Decay	$\alpha$ , $\beta$	4	2+	very high	Helium-4	Helium-4 (stable)
Neutron	$n$	1	0	very high	Neutron	Neutron (stable)
Proton	$p$	1	1+	very high	Proton	Proton (stable)

# Chemactivity 8 Nuclear Chemistry Answers

**Michael P. Garoutte, Ashley B.  
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