

Mole Calculation Worksheet

- 1) How many moles are in 15 grams of lithium?

$$15\text{gLi} \left(\frac{1\text{ mol Li}}{7\text{gLi}} \right) = 2.14\text{ mol Li}$$

- 2) How many grams are in 2.4 moles of sulfur?

$$2.4\text{ mol S} \left(\frac{32\text{gS}}{1\text{ mol S}} \right) = 76.8\text{ gS}$$

- 3) How many moles are in 22 grams of argon?

$$22\text{ gAr} \left(\frac{1\text{ mol Ar}}{40\text{ gAr}} \right) = 0.55\text{ mol Ar}$$

- 4) How many grams are in 88.1 moles of magnesium?

$$88.1\text{ mol Mg} \left(\frac{24\text{gMg}}{1\text{ mol Mg}} \right) = 2,114.4\text{ gMg}$$

- 5) How many moles are in 2.3 grams of phosphorus?

$$2.3\text{gP} \left(\frac{1\text{ mol P}}{31\text{gP}} \right) = 0.1\text{ mol P}$$

- 6) How many grams are in 11.9 moles of chromium?

$$11.9\text{ mol Cr} \left(\frac{52\text{gCr}}{1\text{ mol Cr}} \right) = 618.8\text{ gCr}$$

- 7) How many moles are in 9.8 grams of calcium?

$$9.8\text{gCa} \left(\frac{1\text{ mol Ca}}{52\text{gCa}} \right) = 0.2\text{ mol Ca}$$

- 8) How many grams are in 238 moles of arsenic?

$$238\text{ mol As} \left(\frac{75\text{gAs}}{1\text{ mol As}} \right) = 17,850\text{ gAs}$$

What are the molecular weights of the following compounds?

- 9) NaOH 40 grams

- 12) H_3PO_4 98 grams

- 10) H_2O 18 grams

- 13) Mn_2Se_7 663 grams

- 11) MgCl_2 95 grams

- 14) $(\text{NH}_4)_2\text{SO}_4$ 132 grams

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Julie C. Gilbert



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