Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	lon Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	F.	9	10	-1	Anion
lodine	I-	53	54	-1	Anien
Sulfur	52-	16	18	-2	Anion
Potassium	K+	19	18	+1	Cation
Calcium	Ca ²⁺	20	18	+2	Carlen
Primine	Bc-	35	36	-1	Anien
STEENTINE	Sr2+	38	36	+2	Cation
Oxygen	03-	8	10	-2	Anion
Magnesium	Mart	12	10	+2	Cation
Aluminum	Al 34	13	10	+3	Carion
Selenium	Sea-	34	36	-2	Anion
Lithium	Li*	3	2	+1	Cation
Rubidium	Rb+	37	36	+1	Cation
Chlorine	ce-	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & non metal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.

Mg - =0: => Mg :0:

5. What is the formula for the compound formed in question #4?

Mgo

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M Mark

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at Caltech These collected works by Professor Zewail will certainly be indispensable to both experts and beginners in the field The author is known for his clarity and for his creative and systematic contributions These volumes will be of interest and should prove useful to chemists biologists and physicists As noted by Professor J Manz Berlin and Professor A W Castleman Jr Penn State femtochemistry is yielding exciting new discoveries from analysis to control of chemical reactions with applications in many domains of chemistry and related fields e q physical organic and inorganic chemistry surface science molecular biology etc **Prentice Hall Physical Science Concepts in Action Program Planner National** Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Biological Physics** Philip Nelson, 2013-12-16 Biological Physics focuses on new results in molecular motors self assembly and single molecule manipulation that have revolutionized the field in recent years and integrates these topics with classical results The text also provides foundational material for the emerging field of nanotechnology The Software Encyclopedia 2000 Bowker Editorial Staff, 2000-05 Te HS&T J Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 **Essavs** in **Contemporary Chemistry** Gerhard Quinkert, 2001 In this volume internationally renowned chemists recount their roles in the progress of chemistry research toward elucidation of biological processes Beginning with a historical perspective on the development of X ray crystallography the reader is regaled with first hand accounts of research milestones Included are descriptions of the cutting edge nuclear magnetic resonance and electron paramagnetic resonance spectroscopic techniques the dynamics of ultra fast reactions and the central problem of molecular oxygen activation in biological processes The roles of chiral auxiliaries in organic synthesis and of photochemistry in natural product synthesis are told and innovations in combinatorial synthesis approaches are described Contemporary issues in molecular recognition and modulation of molecular function are addressed concluding with a missive regarding how the frontiers of medical ethics may be breached by molecular manipulations The contributors who number among the finest scientists in the world including two Nobel Prize winners are Peter B Dervan Jack D Dunitz Christian Griesinger Jean Marie Lehn Thomas F Prisner Gerhard Quinkert Peter G Schultz Helmut Schwarz Dieter Seebach and Ahmed Zewail Additionally there is a prologue by Albert Eschenmoser for whom this collection was conceived and an epilogue that contains facsimiles of notes from his landmark lecture Synthesis of Co Enzyme B12 A Vehicle for Teaching Organic Synthesis This book is definitely a must for all who want to read or to read again where we stand in our chemical comprehension of the fascinating relationship between chemical structure and biological processes how we got here and what the future might hold Femtochemistry Ahmed H. Zewail, 1994 Volume II continues with reaction rates the concept of elementary intramolecular vibrational energy redistribution IVR and the phenomena of

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